007) 705/08 AP



Europäisches Patentamt European Patent Office Office européen des brevets

Publication number:

0 337 549

(3)

EUROPEAN PATENT APPLICATION

2 Application number: 89200864.0

(9) Int. Cl.4: C07D 205/08 , A61K 31/395

Date of filing: 06.04.89

Priority: 11.04.88 US 179688

Date of publication of application: 18.10.89 Bulletin 89/42

Designated Contracting States: AT BE CH DE ES FR GB GR IT LI LU NL SE

- (1) Applicant: MERCK & CO. INC. 126, East Lincoln Avenue P.O. Box 2000 Rahway New Jersey 07065-0900(US)
- Inventor: Shah, Shrenik K. 25 Denise Court Metuchen New Jersey 08840(US) Inventor: Finke, Paul E. 34 Inwood Drive Militown New Jersey 08850(US) Inventor: Doherty, James B. 559 Columbia Street New Milford New Jersey 07646(US) Inventor: Barker, Peter L 518 Hort Street Westfield New Jersey 07090(US) Inventor: Hagmann, William 309 Hyslip Avenue Westfield New Jersey 07090(US) Inventor: Dorn, Conrad P. 972 Fernwood Avenue Plainfield New Jersey 07062(US) Inventor: Firestone, Raymond A. 387 Temple Street New Haven Connecticut 06511(US)
- Representative: Hesketh, Alan, Dr. European Patent Department Merck & Co., Inc. Terlings Park Eastwick Road Harlow Essex, CM20 2QR(GB)

[😂] New substituted azetidinones as anti-inflammatory and antidegenerative agents.

Dew substituted azetidinones are found to be potent elastase inhibitors and thereby useful antiinflammatory/antidegenerative agents.

THIS PAGE BLANK (USPTO)

NEW SUBSTITUTED AZETIDINONES AS ANTI-INFLAMMATORY AND ANTIDEGENERATIVE AGENTS

BACKGROUND OF THE INVENTION

We have found that a group of new substituted azetidinones are potent elastase inhibitors and therefore are useful anti-inflammatory/antidegenerative agents.

Proteases from granulocytes and macrophages have been reported to be responsible for the chronic tissue destruction mechanisms associated with inflammation, including rheumatoid arthritis and emphysema. Accordingly, specific and selective inhibitors of these proteases are candidates for potent anti-inflammatory agents useful in the treatment of inflammatory conditions resulting in connective tissue destruction, e.g. rheumatoid arthritis, emphysema, bronchial inflammation, osteoarthritis, spondylitis, lupus, psoriasis, atherosclerosis, sepsis, septicemia, shock, periodontitis, cystic fibrosis and acute respiratory distress syndrome.

The role of proteases from granulocytes, leukocytes or macrophages are related to a rapid series of events which occurs during the progression of an inflammatory condition:

- (1) There is a rapid production of prostaglandins (PG) and related compounds synthesized from arachidonic acid. This PG synthesis has been shown to be inhibited by aspirin-related nonsteroidal anti-inflammatory agents including indomethacin and phenylbutazone. There is some evidence that protease inhibitors prevent PG production;
- (2) There is also a change in vascular permeability which causes a leakage of fluid into the inflamed site and the resulting edema is generally used as a marker for measuring the degree of inflammation. This process has been found to be induced by the proteolytic or peptide cleaving activity of proteases, especially those contained in the granulocyte, and thereby can be inhibited by various synthetic protease inhibitors, for example, N-acyl benzisothiazolones and the respective 1,1-dioxides, Morris Zimmerman et al., J. Biol. Chem., 255, 9848 (1980); and
- (3) There is an appearance and/or presence of lymphoid cells, especially macrophages and polymorphonuclear leukocytes (PMN). It has been known that a variety of proteases are released from the macrophages and PMN, further indicating that the proteases do play an important role in inflammation.

In general, proteases are an important family of enzymes within the peptide bond cleaving enzymes whose members are essential to a variety of normal biological activities, such as digestion, formation and dissolution of blood clots, the formation of active forms of hormones, the immune reaction to foreign cells and organisms, etc., and in pathological conditions such as the degradation of structural proteins at the articular cartilage pannus junction in rheumatoid arthritis etc.

Elastase is one of the proteases. It is an enzyme capable of hydrolyzing the connective tissue component elastin, a property not contained by the bulk of the proteases present in mammals. It acts on a protein's nonterminal bonds which are adjacent to an aliphatic amino acid. Neutrophil elastase is of particular interest because it has the broadest spectrum of activity against natural connective tissue substrates. In particular, the elastase of the granulocyte is important because, as described above, granulocytes participate in acute inflammation and in acute exacerbation of chronic forms of inflammation which characterize many clinically important inflammatory diseases.

Proteases may be inactivated by inhibitors which block the active site of the enzyme by binding tightly thereto. Naturally occurring protease inhibitors form part of the control or defense mechanisms that are crucial to the well-being of an organism. Without these control mechanisms, the proteases would destroy any protein within reach. The naturally occurring enzyme inhibitors have been shown to have appropriate configurations which allow them to bind tightly to the enzyme. This configuration is part of the reason that inhibitors bind to the enzyme so tightly (see Stroud, "A Family of Protein-Cutting Proteins" Sci. Am. July 1974. pp. 74-88). For example, one of the natural inhibitors, α_1 -Antitrypsin, is a glycoprotein contained in human serum that has a wide inhibitory spectrum covering, among other enzymes, elastase both from the pancreas and the PMN. This inhibitor is hydrolyzed by the proteases to form a stable acyl enzyme in which the active site is no longer available. Marked reduction in serum α_1 -antitrypsin, either genetic or due to oxidants, has been associated with pulmonary emphysema which is a disease characterized by a progressive loss of lung elasticity and resulting respiratory difficulty. It has been reported that this loss of lung elasticity is caused by the progressive, uncontrolled proteolysis or destruction of the structure of lung tissue by proteases such as elastase released from leukocytes. J. C. Powers, TIBS, 211 (1976).

Rheumatoid arthritis is characterized by a progressive destruction of articular cartilage both on the free

THIS PAGE BLANK (USPTO)

EP 0 337 549 A1

surface bordering the joint space and at the erosion front built up by synovial tissue toward the cartilage. This destruction process, in turn, is attributed to the protein-cutting enzyme elastase which is a neutral protease present in human granulocytes. This conclusion has been supported by the following observations:

- (1) Recent histochemical investigations showed the accumulation of granulocytes at the cartilage/pannus junction in rheumatoid arthritis; and
- (2) a recent investigation of mechanical behavior of cartilage in response to attack by purified elastase demonstrated the direct participation of granulocyte enzymes, especially elastase, in rheumatoid cartilage destruction. H. Menninger et al., in <u>Biological Functions of Proteinases</u>. H. Holzer and H. Tschesche, eds. Springer-Verlag, Berlin, Heidelberg, New York, pp. 196-206, 1979.

Accordingly, an object of this invention is to discover new protease inhibitors, especially elastase inhibitors, useful for controlling tissue damage and various inflammatory or degenerative conditions mediated by proteases particularly elastase.

Another object of the present invention is to provide pharmaceutical compositions for administering the active substituted azetidinones as protease inhibitors especially human leukocyte elastase.

Still a further object of this invention is to provide a method of controlling inflammatory conditions by administering a sufficient amount of one or more of the active, substituted azetidinones in a mammalian species in need of such treatment.

DETAILED DESCRIPTION OF THE INVENTION

This invention relates to potent elastase inhibitors of formula (I) which are useful in the prevention, control and treatment of inflammatory/degenerative conditions especially arthritis and emphysema.

A large number of the azetidinone derivatives of formula (I) are known antibiotics which have been described in patents and various publications.

The formula of the substituted azetidinones which are found to exhibit anti-inflammatory and antidegenerative activities by the present invention are represented as follows:

$$= \frac{R^{1} R^{2}}{R^{2} R^{2}}$$
 (I)

wherein

20

25

30

35

40

45

55

R can be at the α or the β -position and is hydrogen, straight or branched loweralkyl especially $C_{1-\delta}$ alkyl, such as methyl, ethyl, n- or i-propyl, butyl, pentyl or hexyl; or loweralkyl substituted with a radical R⁴ as defined below; or halo such as fluoro, chloro or bromo:

- R' can be at the α or the β -position and is
 - (1) OB or -S(O)_nB wherein B is as defined below and n is 0. 1 or 2;
- (2) Straight or branched loweralkenyl especially C_{2-3} alkenyl such as vinyl. allyl. $-CH_2CH = C(CH_5)_2$. and $-CH_2CH = CH_2$;
 - (3) loweralkyl as defined above:
- (4) acylamino e.g.

 H H H H

 CHO N -; CF₃CO N -, CH₃CO N -.

or

(5) amino:

- (6) Straight or branched loweralkynyl group especially C₃₋₅alkynyl such as -C=CH. -CH₂-C=CH and -CH₂CH₂-C=CCH₃;
 - (7) An aryl group having 6-14 carbon atoms as described below such as phenyl of formula

5

wherein X₅ and X₆ independently are:

- 1) Q, where Q is H, loweralkyl, haloloweralkyl, phenyl or substituted phenyl as previously defined or naphthyl;
- 15 2) halo.
 - 3) loweralkenyl:
 - 4) loweralkynyi:
 - 5) -SQ:
 - 6) -OQ:
- 20 7) -CHQCOQ1, where Q1 is defined as Q and Q1 can be the same as or different from Q;
 - 8) -CHQCOOQ1 such as CH2COOH;
 - 9) -CH2SQ:
 - 10) -CHQSQ1:
 - 11) -CH2OQ or -CHQOQ1 especially -CH2OH and -CH2OCH3;
- 25 12) -COQ for example, -COCH3 and -(CO)H;
 - 13) -COOQ especially -COOH and COOt-Bu:
 - 14) -OCOQ such as -OCOCH3;
 - 15) -NQQ1:
 - 16) -NQCOQ1 especially -NHCOCH3;
- 30 17) -CH2 NH2 or -CH2N(CH3)317;
 - 18) -CH2OCOCH3;
 - 19) -NQSO2Q1:
 - 20) -SO2NQQ1;
 - 21) -SOQ:
- 35 22) -SO2Q:
 - 23) -SO3Q:
 - 24) -CN:
 - 25) -NO₂;
 - 26) -CONQQ1;
- 40 27) -NO:
 - 28) -CSQ:
 - 29) -CSNQQ';
 - 30) -CF2SQ:
 - 31) -CF2OQ:
- 45 32) -NQCONHQ1 or NQCONQ1Q2 where Q2 is defined as Q1 and Q2 can be the same as or different from
 - 33) -CH₂Y wherein Y represents -CH(NHAC)COO⁺, CH(NH₃) COO⁺, CH₂COOH, COOH, -N(CH₃)₂, OH, CH2N(CH2)2, or CH2OH;
 - 34) -CH(C. -; alkyl);
- 35) -NH(CO)CH2CH2COOH:

36)

55

37) -CO-NH-SOzphenyl or substituted phenyl such as p-chlorophenyl;

5

20

30

40

15

- (d) cycloalkyl having from 3 to 8 carbon atoms especially cyclopropyl, cyclopentyl or cyclohexyl;
- (e) straight or branched alkenyl having from 2 to 20 carbon atoms, for example, allyl;
- (f) straight or branched alkynyl having from 2 to 20 carbon atoms, for example, -C=CH;
- (g) aralkyl, alkaryl, aralkenyl, aralkynyl, alkenylaryl or alkynylaryl wherein alkyl, aryl, alkenyl and alkynyl are as previously defined for example, C₁₋₆ alkylphenyl of formula

$$-(CH2)r \xrightarrow{X_5} or -CH(C1-6alky1)(CH2)r-1 \xrightarrow{X_5} X_6$$

wherein r is 1 to 6. C+=salkyl naphthyl of formula

0t 12

55

50

(h) heteroaryl comprising monoheteroaryl, di- or polyheteroaryl, or fused heteroaryl containing from 1 to 3 of any one or more of the heteroatoms N, S or O in each heteroaryl ring thereof, for example, pyridyl, pyrryl, such as

- 8) heteroaryl such as unsubstituted or substituted furyl, thienyl, thiazolyl, pyrryl, pyrimidinyl, pyridyl, oxazolyl, tetrazolyl or imidazolyl wherein the substituents are as those described for substituted phenyls:
 - 9) aralkyl especially phenyl C1-6alkyl such as benzyl of formula

10

15

or

20

25

30

40

45

50

or phenethyl;

- (10) halo such as F. Cl, Br or I;
- (11) N₃;
- (12) hydrogen:
- (13) R and R¹ may join together and form a cycloalkyl such as a C1-scycloalkyl, e.g., cyclopentane. = $C(B)(B_1)$ or = O(0x0) wherein B and B_1 independently are as defined below:
 - (14) -CH₂OC₁₋₅ alkyl especially -CH₂OCH₃ and -CH₂OC₂H₅;
 - (15) -CH₂CH₂OC₁ = alkyl especially -CH₂CH₂OC₂H₅;
- can be at the α or the β -position and independently are R² and R³
 - (1) B as defined below:
 - (2) -CONBB+ wherein B and B+ independently represent
- (a) H;
- (b) straight or branched alkyl having from 1 to 20 carbon atoms, preferrably C--salkyl such as methyl. ethyl, isopropyl, t-butyl, pentyl or hexyl;
- (c) aryl having from 6 to 14 carbon atoms such as phenyl or substituted phenyl of formula



naphthyl or substituted naphthyl of formula

or anthracyl or substituted anthracyl of formula 55

thienyl, isothiazolyl, imidazolyl such as

10

15

5

tetrazolyl such as

20

25

30

35

÷0

45

50

$$N = N - R^{5}$$
or
$$N = N$$

$$N = N$$

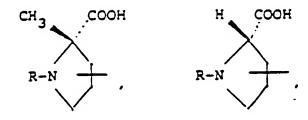
$$CH_{2} CC_{6}^{H_{5}}$$

pyrazinyl, pyrimidyl, quinolyl, isoquinolyl, tetrahydroisoquinolyl such as

benzothienyl, benzofuryl such as

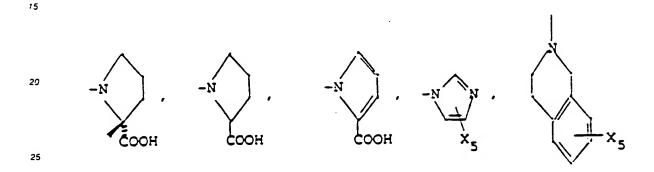
pyrazolyl, indolyl, purinyl, carbazolyl, isoxazolyl and the like:

- (i) heteroarylalkyl such as 2-pyridylmethyl, 2-thienylmethyl and 3-isothiazolylethyl; or
- (j) heterocycloalkyl e.g., 1.3-dioxacyclohex-4-yl, piperidino,



morpholino, oxacyclopropyl, pyrrolidino, benzothiazolino, imidazolidino, pyrazolidino, and piperazino:

- (k) heterocycloalkenyl such as pyrrolino, 2-imidazolino, 3-pyrazolino or isoindolino;
- (I) B and B_1 joined together and form a heterocyclic ring containing at least one N-atom and optionally 1 to 3 of the heteroatoms, N, S, or O, e.g.,



the above groups (a)-(I) can be unsubstituted or can be substituted by one or more radical R² selected from the group consisting of loweralkyl, hydroxy, aryloxy (OAr), alkoxy, halo, nitro, loweralkylthio, arylthio, mercapto, amino, monoalkyl or dialkyl substituted amino, cyano, carboxy, loweralkanoyl, Ar(C = O), aminosulfonyl, aminosulfenyl, aminosulfinyl, carbamoyl, carbamoyloxy, -S(O)_nR⁵, SO₃R⁵, -P(O)_qR⁵ (where q is 1 or 2 and R⁵ is H, C₁₋₅ alkyl, aralkyl or aryl as previously defined), azido, carboxamido or N-substituted carboxamido;

 $(3) -S(O)_nB$;

35

40

45

50

5

- (4) -S(O), NBB1;
- (5) -N(B)S(O),B1:
- (6) -P(O)_qBB₁;
- (7) -C(O)B especially acetyl, benzoyl, e.g., p-chlorobenzoyl, p-methylbenzoyl and p-aminosulfonyl-benzoyl;
 - (8) -OB especially -OC: -6alkyl, phenoxy or substituted phenoxy of formula

$$-0$$
 \times_5
and
 \times_5

(9) -COOB, -OC(O)OB or OC(O)NBB+;

- (10) -O-C(O)-B especially C_{1-5} alkanoyloxy such as acetyloxy:
- (11) cyano;
- (12) -S-C(O)-B especially C1-6 alkanoylthio such as acetylthio; or
- (13) R^2 and R^3 may join and form = $C(B_1)(B)$, a C_1 -cycloalkyl for example, cyclopentyl, and = 0-(OXO):

55 A is

- (1) -O-C(O)-B;
- (2) -S(O)_nB:
- (3) -S(O),NBB1;

(4) -C(O)B:

(5) SO₃-M° wherein M represents

(a) an alkali anion such as Na . K ; or

(b) a quaternary ammonium group of formula N°(R5)4, for example, (n-Bu)4N°;

(6) substituted or unsubstituted phosphoryl or phosphonyl such as $-P(O)_1(R^5)_2$ or $-P(O)_4R^5$;

(7) -C(O)NBB₁ especially -CON(C₂H₅)Phenyl and -CONHB₁, wherein B₁ is

(a) -(CH₂), Ph. or CH(C₁- $_{6}$ alkyl)(CH₂),.1Ph. e.g.. -CH(C₂H₅)-Ph. -CH(C₃H₇)-Ph. -CH-(allyl)-Ph. -CH(C₂H₅)-Ph. -CH(C₃H₇)-Ph. -CH-(allyl)-Ph. -CH-(C₂H₅)-Ph. -CH-(allyl)-Ph. -CH-(C₂H₅)-Ph. -CH-(C₃H₇)-Ph. -CH-(allyl)-Ph. -CH-(C₃H₇)-Ph. -CH-(C₃H₇)-Ph. -CH-(allyl)-Ph. -CH-(C₃H₇)-Ph. -CH-(C₃ CH₂Ph, or -CH(CH₃)-Ph wherein Ph represents phenyl or substituted phenyl as previously defined, for examples, 4-methyl-phenyl, 4-methoxyphenyl, 4-N,N-dimethyl-amino-phenyl, 4-benzyloxy-phenyl, 4-phenylphenyl, 3,4-methylenedioxy-phenyl, and 3,4-dimethyl-phenyl;

(b) -(CH₂),(Naph) especially -CH₂(Naph) or -CH(C₂H₅)(Naph) wherein (Naph) is α or β-naphthyl or substi-

tuted naphthyl as previously defined;

(c) -(CH₂)_r(Ar) especially -CH₂CH₂Ar or -CH₂Ar wherein Ar represents heteroaryl especially 2-thienyl. 2furyl, 3-thienyl, or benzofuryl;

15 (d) -(CH₂),OPh especially -CH₂CH₂CH₂OPh;

(e) -(CH₂),CH(OH)Ph;

(f) -(CH₂),(CO)Ph:

(g)

20

25

5

30 (h)

35

40

(i) -CH2-Ph(p-CO-NH-SO2-Ph(p-Cl));

(j)

45 (k)

55

50

or

(I) -CH(CH₃)-Ph(p-cyclohexyl);

- (8) -C(O)OB especially C_{1-6} alkoxycarbonyl, e.g., methoxycarbonyl, and -ethoxycarbonyl or C OOCH₂-Ph(p-COOC₂H₅);
 - (9) halo C: -s alkyl such as trifluoromethyl;
- (10) -OB especially -O-CH₂-(phenyl or substituted phenyl as previously defined), for example, -OCH₂C₆H₅; -OCH₂C₆H₄-OCH₃; or OCH₂C₆H₄NO₂;
 - (11) silyl such as -Si(CH₃)₂(t-Bu):
- (12) B especially H, C₁₋₅ alkyl, CH₂OH, -CH₂O(CO)CH₃, phenyl or substituted phenyl, -CHR⁵L where R⁵ is as previously defined and L is a good leaving group comprising OAc, SAc, halogen, OR⁵, SCR⁵, SO₂R⁵, OTs, OCOCF₃, and mesyl wherein Ac is acetyl; and Ts is tosyl; or

(13)

Preferably, the compounds of the present invention are of formula (I) wherein:

20 R is hydrogen; or C₁₋₆ alkyl;

R! is

- (1) C: -s alkyl especially methyl or ethyl
- (2) OR5:
- (3) phenyl of formula

25

5

15

$$X_{x_{\epsilon}}$$

30

- (4) hydrogen:
- (5) benzyl of formula

35

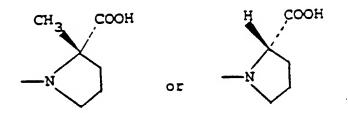
40

55

- (6) CH2OC+=5 alkyl or -CH2CH2OC+=5 alkyl;
- (7) C_{2-8} alkenyl: or
- (8) R and R¹ may join together and form a cyclopentane:

R² and R³ independently are

- (1) hydrogen;
- (2) S(O),R5;
- (3) COOB:
- 50 (4) CONBB:;
 - (5) OB;
 - (6) C1-5 alkyl;
 - (7) phenyl or substituted phenyl as previously defined:
 - (8) naphthyl as previously defined;
 - (9) cyclohexyl;
 - (10) benzyl as previously defined:
 - (11) heteroaryl selected from a group consisting of imidazolyl, benzofuryl and tetrahydroisoquinolyl;
 - (12) heterocycloalkyl such as



(13) C(O)B; 10

is

5

- (1) SOR5;
- (2) SO₂R⁵;
- (3) COOB:
- (4) C(O)B; 15
 - (5) CONBB, wherein B and B, independently are;
 - (a) H:

25

30

40

45

- (b) C: -salkyl;
- (c) -(CH₂), Ph where r represents 1 or 2 and Ph represents phenyl or substituted phenyl as previously defined;
 - (d) CH(C₁₋₅ alkyl)(CH₂)_{r-1}Ph where Ph is as defined above;
 - (6) phenyl or substituted phenyl as previously defined such as p-methoxyphenyl, p-nitrophenyl and p-methylphenyl;
 - (7) C_{1-5} alkyl;
 - (8) CH2OH;
 - (9) CH2OC(O)CH3; or
 - (10) -C(O)NHSO2-Ph(p-CH3).

Even more preferably, the compounds of the present invention are of formula (I) wherein

- is hydrogen or C1-3alkyl: R
- R.
 - (1) hydrogen:
 - (2) C. -; alkyl:
 - (3) C₁₋₅ alkoxy such as methoxy;
- (4) C2-5 alkenyl; 35
 - (5) phenyl or substituted phenyl as previously defined; or
 - (6) CH2OC1-3 alkyl;
 - R2 is hydrogen;
 - R² is
 - (1) S(O),R5;
 - (2) CONBB1;
 - (3) COOB:
 - (4) phenoxy or substituted phenoxy;
 - (5) imidazolyl; or
 - (6) substituted or unsubstituted alkoxy, for examples, OCH₃ or -OCH₂CONH₂;

- (1) CO_2R^5 wherein R^5 is H. C_{1-5} alkyl. -CH₂Ph. -CH(CH₂)Ph. -CH(C₂H₅)Ph. CH(C₃H₇)Ph or CH-(C3H5)Ph wherein Ph represents phenyl or substituted phenyl as previously defined:
 - (2) CONHR5; or
- (3) SO₂R⁵. 50

The most preferred compounds of the present invention are listed in the following table.

1	υ	

20

5

	R	R۱	R ⁵
	C ₂ H ₅	CH₃	CH(CH₃)Ph
	C₂H₅	C₂H₅	CH ₂ -(4-Ph-Ph)
1	C ₂ H ₅	CH₃	CH(C₂H₃)Ph
1	C₂H₅	C₂H₅	CH(C₂H₅)Ph
1	C ₂ H ₅	CH ₂ OCH ₃	CH(C₂H₅)Ph
ł	C ₂ H ₅	C ₂ H ₅	CH(C₃H₂)-(4-CH₃-Ph)
	C ₂ H ₅	C₂H₅	CH(C ₃ H ₅)-(4-CH ₃ -Ph)
	C₂H₅	C₂H₅	CH(C₃H₂)Ph
	C ₂ H ₅	C₂H₅	CH(C₃H₂)-(3,4-methylenedioxy-Ph)
	C ₂ H ₅	C₂H₅	CH(C₃H₅)-(3,4-methylenedioxy-Ph)

In the above table. Ph represents phenyl or substituted phenyl as previously defined.

The compounds of the present invention are either known or are prepared among other methods by the following representative schemes.

30 Scheme (a)

as illustrated by Examples 16-19.

55 wherein

Y is -NO₂, -CH₃, -OCH₃, -Cl, -F, etc;

X is halo, e.g., Cl, Br or I;

Z is BCO or BSO₂.

Scheme (b)

as illustrated by Examples 1-4.

5

10

wherein

X is halo:

Z is as previously defined, e.g., -SO₂-(p-NO₂-Ph), -COCH₃, -CH₂OTs, etc. wherein Ph represents phenyl or substituted phenyl.

Scheme(c)

as illustrated by Examples 5-15

25

20

35

30

40

45 wherein

R6 is H. CF3, CH3, etc.;

R5 and R1 are as previously defined; and

CAN is cerric ammonium nitrate.

50 Scheme (d)

as illustrated by Examples 2-3.

Scheme (e)

5

10

as taught by M. A. Krook and M. J. Miller (J. Org. Chem., 1985, 50, 1126-1128), the following type of compounds can be prepared.

Scheme (f)

30

as taught by Hart, D. J. et al.. (J. Org. Chem., 48, pp. 289-294, 1983); the following class of compounds can be prepared.

$$R^{2}CHO + LIN(TMS) \longrightarrow R^{2}CH = N(TMS)$$

$$R^{1} CHCOOR^{5} \longrightarrow R^{2}C = C OLi$$

$$R^{1} CHCOOR^{5} \longrightarrow R^{2}C = C OLi$$

wherein ${\sf R}^{\sf 5}$ is as previously defined: and TMS is trimethylsilyl.

Scheme (g)

as taught by P. J. Reider and E. J. J. Grabowski (Tet. Lett., 23, p. 2293, 1982); the following groups of compounds can be prepared.

55

45

wherein R1 is as previously defined.

Scheme (h)

15

as illustrated by Examples 20 and 21:

This invention also relates to a method of treating inflammation in patients using a compound of Formula (I), particularly a preferred compound as the active constituent.

It has been found that the compounds of Formula (I) are effective inhibitors of the proteolytic function of human granulocyte elastase as shown below:

55

TABLE I

5	R ¹ R ² H
10	

 $^{\text{ID}}_{50}$ $^{\text{Ki}}_{\text{obs}}^{\text{/I}}_{1}$ $^{\text{obs}}_{1}$ $^{\text{H}}_{1}$ $^{\text{H}}_{2}$ $^{\text{Rec}}_{2}$ R² 15 10.00 COCH₃ SOCH3 Н 20 3.00 COCH OCOCH н 15.00 ососн 15100 0.10 0.36 OCOCH₃ COCH COCH 0.01 25 OCOCH n-propyl 10.00 C₆H₅ (trans) COOC 2H5 $50_{2}(p-C_{6}H_{4}-NO_{2})$ 3.00 COOCH₂C₆H₅ н 0.50 COCH₃ СНЗ OCOCH₃ CH 30 4.00 $50_{2}(p-C_{6}H_{4}-NO_{2})$ C00C2H5 C6H5 (trans) 3.00 $50_2(p-C_6H_4-NO_2)$ C00C2H5 C6H5 (cis) 2.00 COCH COOCH₂C₆H₅ CH₃0 35 503(Bu)4N+ 8.00 OCOCH₃ n-propyl 50₂(p-C₆H₄-NO₂) 0.02 COOC₂H₅ C2H3 (cis) 3925 0.05 $50_{2}(p-C_{6}H_{4}-NO_{2})$ C2H5 (cis) COOC2H5 39300 0.05 $50_{2}(p-C_{6}H_{4}-NO_{2})$ 40 COOC₂H₅ C2H5 (trans) $50_{2}(p-C_{6}H_{4}-CH_{3})$ 0.01 C₂H₅ (trans) C00C2H5 0.06 $50_2(p-C_6H_4-NO_2)$ C00C2H5 n-propyl (trans) 0.05 $50_2(p-C_6H_4-NO_2)$ CH₃CHCH (cis) C00C2H5 **4**5 1.50 $p-(C_6H_4-NO_2)$ CH₂CH 4514 2.00 COCH₃ OCOCH2CH2COOH C2H5 310C0 0.19 COCH OCOPh C2H5 (trans) 50

EP 0 337 549 A1

<u>TABLE I</u> (Continued)

R		R ¹	R ²	A	(ha\wj) ₁₀ 20	Ki (μπ)	k /I obs /I (H sec 1)
-					•		
н	•	C2H5 (cis)	OCOPh	COCH ₃ -		0.21	28500
н		C ₂ H ₅	ососн	COCH ₂ CH ₂ COOH		1.43	2250
н		C ₂ H ₅ (cis)	ососн	COPh		0.14	
H		C ₂ H ₂ (trans)	ососн	COPh		0.34	76600
H		C _H (trans)	OPh	COCH ₃		4.30	5270
		⁻² 5 C ₂ H ₅ (trans)	0C ₂ H ₅	COCH ₃		11.90	1670
		2 5 (trans) 2 5	OPh-p-COOH	COCH ₃		3.40	8727
		C ₂ H ₅ (trans)	OPh-p-COOH	COOC ₂ H ₅		2.10	8680
	H	2 5 'C _H (trans)	OPh-p-COOH	CONHCH ₃		16.50	
		C ₂ H ₂ (cis)	CON(CH ₂) ₄	502(P-C6H4-CH3)	27.70	541
	н	C ₂ H ₅ (cis)	COOCH ₂ C ₆ H ₄ -p-COOH	•		4.20	299
	н	C ₂ H ₅ (cis)	CON(CH ₃)CH ₂ COOH	50 ₂ (p-C ₆ H ₄ -CH ₃		22.00	165
	н	C _H (trans)	осн ₂ соон	COOC2H5		-	512
	н	² 5 C ₂ H ₅ (cis)	осн ₂ соон	C00C2H5		-	796
	н	n-propyl (trans)	_	C90C2H5		-	1504
	н	C ₂ H ₅ (trans)	осн ₂ соинсн ₂ соон	COOC ₂ H ₅		-	1000
	н	C ₂ H ₅ (trans)	осн(сн ₃)соон	C00C2H5		-	346
	н	2 5 (cis) C ₂ H ₅ (cis)	соосн ² соон	50 ₂ (P-C ₆ H ₄ -CH ₃	,	-	1554

ID is the effective dosage in micrograms per milliliter (μ g/ml) for 50% inhibition of the enzyme activity two minutes after time zero. Ki is the concentration of the inhibitor (micromolar, μ H) giving 50% of the control enzyme activity. k /I (M l sec $^{-1}$) is the second order rate constant obs

Table II

R CO-NH-E

10			•		
	R	R ¹	R ²	81	<u> </u>
		СН3	0-(4-COOH-Ph)	CH ₂ Ph	1900
15	с ₃ н ₇	CH ₃	0-(4-COOH-Ph)	CH(CH ₃)Ph	15,000
.5	с ₂ н ₅	3 Н	0-(4-COOH-Ph)	CH ₂ Ph	5,000
	^С 3 ^Н 7		0-(4-CO(CH ₂)2COOH-Ph)	CH ₂ (4-Ph-Ph)	107,045
	^C 2 ^H 5	С ₂ Н ₅	0-(4-COOH-Ph)	CH ₂ (4-Ph-Ph)	37,000
20	C ₂ H ₅	с ₂ н ₅ сн осн	0-(4-COOH-Ph)	CH ₂ (4-Ph-Ph)	44,533
	^C 2 ^H 5	сн ₂ осн ₃ с ₂ н ₅	0-(4-NO ₂ -Ph)	CH ₂ Ph	6.947
25	C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ (2-Anthracene)	36.177
	C H	сн	0-(2-CH ₂ CH-Ph)	CH ₂ Ph	2961
30	с ₂ н ₅ с ₂ н ₅	C ₂ H ₅ C ₂ H ₅	0-(4-CH ₂ COOH-Ph)	CH ₂ Ph	3175
	^С 2 ^Ч 5	c ₂ H ₅	0-(4-CH ₂ CH-NH ₃ -Ph) co ₂	CH ₂ Ph	2540
35	C ₂ H ₅	C2H5	0-(4-NHCOCH3-Ph)	CH ₂ Ph -	3503
	C ₂ H ₅	C ₂ H ₅	0-(4-NHCOCH2CH2COOH-Ph)	CH ₂ Ph	2568
	C ₂ H ₅	C ₂ H ₅	0-(4-CH ₃ CO-Ph)	CH ₂ -(4-COOH-Ph)	2307
40	2 5 C ₂ H ₅	C ₂ H ₅	0-(4-CH ₃ CO-Ph)	CH ₂ (4-CH ₃ CO-Ph)	5916
	2 5 C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ -(2-fury1) -	5223
	C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ -(2-thienyl)	4925
45	C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	n-C9 ^H 19	8300
	2 5 C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	(CH ₂) ₃ Ph	4537
	C _z H _s	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ Naph	21,269

55

50

EP 0 337 549 A1

	_	R ¹	R ²	8,	<u> </u>
	3		0-(4-COOH-Ph)	(CH ₂) ₄ Ph	10,894
	C _Z H _S	С ₂ Н ₅	0-Ph	CH ₂ -(4-COCH-Ph)	1501
	C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ -cyclohexyl	1424
	^C 2 ^H 5	^C 2 ^H 5	0-(4-C00H-Ph)	CH ₂ Ph _	4000
	^C 2 ^H 5	Н	0-(4-COOH-Ph)	CH ₂ Ph	2000
	с ₂ н ₅ сн ₂ сн - сн ₂	сн ₃ н	0-(4-COOH-Ph)	CH ₂ Ph	5400
	C ₃ H ₇	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ Ph	3230
	cyclopentane		0-(4-COOH-Ph)	CH ₂ Ph	1900
	(R and R comb				
	form the cyclo	pentane ring)			
	С ₂ Н ₅	сн ₂ осн ₃	0-(4-COOH-Ph)	CH ₂ Ph	1900
	C ₂ H ₅	CH ₃	0-(4-COOH-Ph)	CH2CH(CH3)Ph	2553
	2 5 C ₂ H ₅	C ₃ H ₇	0-(4-COOH-Ph)	CH ₂ -(2-Naph)	51,000
	C ₂ H ₅	C ₂ H ₅	0-(4-CDOH-Ph)	CH(CH ₃)-(1-Naph)	14,128
	² 5 C ₂ H ₅	C ₂ H ₅	0-(4-CDOH-Ph)	CH ₂ -(4-C1-Ph)	3419
	C ₂ H ₅	2 5 C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ (4-CH ₃ -Ph)	3965
		2 5 C ₂ H ₅	0-(4-C00H-Ph)	CH ₂ (4-F-Ph)	2337
	С ₂ Н ₅ С Н	2 5 C ₂ H ₅	0-(4-C00H-Ph)	CH ₂ (4-0CH ₃ -2h)	5162
	с ₂ н ₅ с н	2 5 C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ (4-NO ₂ -Ph)	5075
	С ₂ Н ₅ С Н	C ₂ H ₅	0-(4-C00H-Ph)	CH(CH ₃)-(3-C1-4-c	yclo-
	^C 2 ^H 5	2 5		hexyl-Ph)	20,776
	СН	сн ₂ осн ₃	0-(4-COOH-Ph)	CH ₂ -(3,4-methylen	e- <i>*</i>
)	^C 2 ^H 5	2 3		dioxy-Ph)	16.984
	СН	с ₂ н ₅	0-(4-COOH-Ph)	CH ₂ -(2-benzofuran	13.15
	с ₂ н ₅ с н	C ₂ H ₅	0-(2-(6-COOH-Naph))	CH ₂ Ph	5561
5	с ₂ н ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ (4-(4-C1-Ph)-	
	^C 2 ^H 5	2 5		SO2NHCO-Ph)	1730

	R	g l	R ²	8,	<u> </u>
5 .	C ₂ H ₅	C ₂ H ₅	:-(3-C0-NHCH ₂ -Ph) ССООН	CH ₂ Ph	3047
J .	C ₂ H ₅	с ₂ н ₅	0-(3-COOH-Ph)	CH ₂ Ph	1763
	2 5 C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	CH ₂ -(4-Ph0-Ph)	12.036
10	c ₂ H ₅	[©] z ^H 5	0-(4-COOH-Ph)	CH ₂ -(4-HN(CH ₃) ₂ -Ph) •CF ₃ COO	9983
	с ₂ н ₅	с ₂ н ₅	0-(4-COOH-Ph)	(CH ₂) ₄ 0Ph	3447
15	C ₂ H ₅	C ₂ H ₅	0-(4-COOH-Ph)	(CH ₂) ₄ CH(GH)Ph	4200
	C ₂ H ₅	с ₂ н ₅	0-(4-N(CH ₃) ₃ I-Ph)	CH ₂ Ph	1700
20	C ₂ H ₅	с ₂ н ₅	-1-imidazolyl	CH ₂ Ph	200
25	с ₂ н ₅	с ₂ н ₅	О	CH ₂ Ph	2000
30					
35	^с 2 ^н 5	^C 2 ^H 5	COHN-SO2	CH ₂ Ph	5300
10	с ₂ н ₅	^С 2 ^Н 5	мнсосн 3	CH ₂ Ph	2422

EP 0 337 549 A1

Table II
(continued)

		R1	R ²		k _{obi} /I
5	C H	Н	0-(4-C00H-Ph)	Ph-4-COOH	13,563
	С ₂ Н ₅ С Н	c ₃ H ₇	0-(4-COOH-Ph)	CH ₂ Ph	2,500
	C ₃ H ₇ allyl	C ₂ H ₅	0-(4-COCH-Ph)	CH ₂ Ph	1974
10		C ₂ H ₅	O-(4-COOH-Ph)	CH ₂ Ph	87
	CH ₂ Ph C H	-2-5 сн ₂ осн ₃	0-(4-COOH-Ph)	CH ₂ -2-Naph	50,000
	С ₂ Н ₅	H 3	Ph-4-COOH	CH ₂ Ph	900
	С ₂ Н ₅ н	OMe	Ph-4-COOH	CH ₂ -2-Naph	1340
15		с ₃ н ₇	0-(4-COOH-Ph)	CH ₂ Ph-3-CF ₃	55,000
	^С 2 ^Н 5 С Н	СН ₃	0-(4-COOH-Ph)	CH(Et)-5-benzofury1	750.000
	С ₂ Н ₅ С н	CH ₃	0-(-4-COOH-Ph)	CH(Et)-3-thienyl	78.800
20	С ₂ Н ₅ С Н	з сн ₂ осн ₃	0-(4-COOH-Ph)	CH(nPr)Ph	75.000
	С ₂ Н ₅ С Н	C ₃ H ₇	0-(4-	CH(Et)Ph	87.000
	^C 2 ^H 5	3 7	CO(CH ₂) ₂ COOH_Ph)		
25	СН	с ₃ н ₇	0-(4-CH ₂ COOH-Ph)	CH(Et)Ph	54,000
	с ₂ н ₅	3 7 CH ₃	0-(4-COOH-Ph)	Cyclopentyl	-
	С ₂ Н ₅	3 CH ₃	0-(4-C00H-Ph)	CH(CH ₃)CH ₂ CH ₂ CH ₃	
	С ₂ Н ₅	сн ₃	0-(4-CONH ₂ Ph)	CH ₂ Ph	12,500
30	С ₂ Н ₅	CH ³	0-(4-COOH-Ph)	CH ₂ (3,5—diMe—	
	^C 2 ^H 5	3		4-C00H-Ph)	5.500
	ĊН ·	сн _з	0-(4-CONH ₂ Ph)	CH ₂ (3.5-diMe-	
35	^C 2 ^H 5	3	4	4-COOH-Ph)	30,000
-	C2H5	CH ₃	0-(4-COOH-Ph)	CH ₂ (3.4-diMeO-Ph)	11,300

40

20

He represents CH₃

Ph represents phenyl

Pr represents propyl

Bu represents butyl

Table III

C 2 H 5 CONHB₁

5

10

20

25

30

35

40

45

50

55

CH₂Ph-4-Ph 2901 OCH₂COOH 15

> 4157 0-(4-COOH-Ph)

CH2Ph-4-Ph 12,545 0-(a11y1)461 CH2Ph-4-Ph -1-imidazolyl CH2Ph-4-Ph 2144 1-triazolyl 3658

CH₂Ph (1-methyl-tetrazol-5-yl)thio 116 CH₂Ph (1-H-triazol-3-yl)thio

948 CH₂Ph 1-tetrazolyl 357 CH, Ph [2H-1-pyridony1]

CH₂-2-Naph(6-C00H) 0-Ph-4-CONH₂ 40,650 69

CH₂Ph 1-benzimidazolyl

351 CH₂Ph

818 CH₂Ph 0-glyceryl 51,802 CH2-Ph-4-Ph OCH2CONH2

496 CH₂Ph NH-COOMe

5711 CH-(Et)-Ph осн₂-соон 102,974

CH-(Et)-Ph OCH2-CONH2 nBu

0-(4-COOH-Ph) cyclopentyl 0-(4-COOH-Ph)

O-CH2CON(Et)2 CH(Et)Ph

EP 0 337 549 A1

Table III (Continued)

5	R ²		k _{obs} ∕I .
	0-(4-COOH-Ph)	CH ₂ Ph(2-OH)	1461
	0-(4-COOH-Ph)	CH ₂ Ph(4-tBu)	21,774
10	0-(4-COOH-Ph)	CH ₂ Ph(4-(3-COOH)Ph)	14,727
	0-(4-COOH-Ph)	CH ₂ Ph(4-CO-HCD)	2036
15	0-(4-COOH-Ph)	CH ₂ Ph(4-CH ₂ Ph)	8032
	0-(4-COOH-Ph)	CH2Ph(3-CH3)	6932
	0-(4-COOH-Ph)	CH ₂ Ph(3,4-(CH ₂) ₄)	62,883
20	0-(4-COOH-Ph)	CH ₂ Ph(3,4-0iMe)	20,600
	0-(4-COOH-Ph)	CH ₂ Ph(4-i-Pr)	18,846
	0-(4-COOH-Ph	CH ₂ Ph(4-S(0) ₂ He)	3350
25	0-(4-COOH-Ph)	CH ₂ Ph(4-COMe)	5916
	0-(4-COOH-Ph)	CH ₂ Ph(4-OMe-3-Me)	13', 126
	0-(4-COOH-Ph)	CH2-Ph(4-0CH2Ph)-	12.036
	0-(4-CH(COOH)NHAc-Ph)	CH ₂ Ph	1675
30	0-(4-CH(0H)COOH-Ph)	CH ₂ Ph(3,4-0iMe)	17,525
	0-(3-0H-4-C00H-Ph)	CH ₂ Ph(4—Me)	9252
	0-(2-(CH ₂)3NMe2-Ph)	CH ₂ Ph	629
35	0-(4-CH ₂ COOH-Ph)	CH ₂ Ph(4-Ph)	28,870

Table IV

5 СОВ 10 0-(4-CCOH-Ph) -СH₃ 15 10.066 0-(4-COCH-Ph) 20 1446 (lower ref.) 0-(4-COCH-Ph) C3H7 25 5977 -N(CH2Ph)2 0-(4-COOH-Ph) -OCH2-(4-COOC2H5-Ph) 0-(4-C00H-Ph) 30 -OCH2-(4-COOC245-Ph) 14,331 0-(4-COOH-Ph) 82.956 -N(C2H5)(CH2Ph) 35 0-(4-C00H-Ph)

40

45

50

EP 0 337 549 A1

Table V

x₅

	X	<u>H</u>	x ₆	<u> </u>
	4-COOH	Et	• н	92.000
20	4-C00H	nPr	н	152.000
	4-COOH	CH ₂ 0Me	н	6.094
	4 —сн ₂ соон	Et	н	140,000
25	4-COOH	Me	4-He	47,000
	4-C00H	Et	4-Me	_
	4-COOH	PhCH ₂	н	25,000
30	4—СН ₂ СООН	nPr	н	227,000
	4-COOH	nPr	CH ₃	
	4-C00H	nPr	н	120,000
	4-C00H	Et	3,4-(0CH ₂ 0)	
35	4-CH ₂ COOH	nBu	н	
	4-C00H	allyl	н	

Table VI

5	n x,	
•		
10	ĭ	•
70	ਜ -	

	X ₅	М	x ₆	k _{obs} /I
15	4-C00H	He ·	н	4016
	4-C00H	Me	4-Ph	74,000
	4-CH ₂ COOH	Me	н	8.373
20	4-C00H	Me(s)	4-Ph	49246
	4-COOH	Ph	Ph	67754
	4-C00H	Me	4-(2'-C1-Ph)	245130
25	4-COOH	Et	4-Ph	26382
	4-CCOH	Et	н	76204
	4-CO-(CH ₂) ₂ -COOH	Me	н	37084
20	4-CO-(CH ₂) ₂ COCH	Et	н	272190
30	4-C00H	nPr	н	116060
	3.5-Me ₂ -4-CCOH	Et	н	24,994
	4CH ₂ COOH	Et	н	126,000
35	3-0H-4-CCOH	Et	н	-124560
	3-CH ₂ COOH	Me	н	5885
	4-CH-CH-COOH	Me	н	9101
-10	4-C00H	CH ₂ OMe(S)	н	6981
	4-CH ₂ COOH	CH ₂ OMe(S)	н	
	4-COOH	Me	-He	10680
	4-COOH	iPr(S)	н	4743
45	4-COOH	iPr	н	177075
	4-сн ₂ соон	nPr	н	188,000
	4-CH ₂ COOH	CH ₂ OMe(R)	Н	11004
50	3,5-He ₂ -4-COOH	nPr	н	

Table VI (Continued)

	x ₅	М	x ₆	<u>k_{obs}/!</u>
-	3-CH ₂ COOH	Et	4—He	
10	4-(CH ₂) ₂ COOH	Me	н	9481
	3-CH ₂ COOH	Et	н	81018
	4-C00H	CH ₂ OMe(R)	н	6981
15	4-C00H	Et	3 <i>-</i> He	
	4-CH ₂ COOH	Et	3-He	
	4-C0(CH ₂) ₂ C00H	allyl	4-Me	
20	4-C00H	Me	4-He	
	4-CH ₂ COOH	Et	3-01	
	4-C00H	Et	3-01	
	4C00H	allyl	3-He	
25	4-COOH	nPr	3–Me	
	4CH ₂ COOH	allyl	4-Me	664,000
	з-сн ₂ соон	allyl	4-Me	
30	4-CH ₂ COOH	allyl	3 <i>-</i> Me	
	4CH ₂ COOH	nPr	3-Me	
	4-CO(CH ₂) ₂ COOH	nPr	4-He	
35	3-CH ₂ COOH	allyl	н	• •-
	3-сн ₂ соон	CH ₂ OMe(S)	н	
	4-COCH	allyl	н	•
	4-CH ₂ COOH	allyl .	н	
40	4-COOH	Et	4-He	
	4-COCH	Et(S)	4-He	
	4-COOH	allyl	4-Me	
45	4-C00H	nPr	4–He	389,000
	3-CH ₂ COOH	nPr	4 -M e	
	4-CH ₂ COOH	. nPr	4-Me	557,000

50

Table VI (Continued)

5		X	. н		<u>х</u> _б	<u>k</u> obs
		3-CH2COOH	Et		4-C1	•
10		2 4–C00H	Et		4-01	
		4-СН ₂ СООН	Et		4-He	
		3-CH ₂ C00H	Et		3-01	
		4-C00H	allyl		3,4—methylenedioxy	-
15		4-C00H	nPr		3,4—methylenedioxy	
		4_CH ₂ COOH	allyl		3,4—methylenedioxy	605.000
		4_CH ₂ COOH	nPr		3,4-methylenedioxy	857.000
20		3-CH ₂ COOH	сн ₂ соон		- 4-He	
		3-CH ₂ COOH ·	nPr		н	
		4—C00H	Et		3.4-methylenedioxy	
		4-CH ₂ C00H	Et		3,4—methylenedioxy	
25	-	4-C00H	Et		3,4-He ₂	
		4-C00H	сн ₂ с≡ссн ₃	н		
		4-CH ₂ COOH	сн ₂ с≣ссн ₃	н		
30		4-C00H	nBu		н	
		2-NO ₂ -4-CH ₂ COOH	Et		н	
		4—СООН	Et	-	4 - F	
		4C00H	Et		3-Me-4-0Me	
35						••

Protocol - Enzyme Assays for the Inhibition of Human Polymorphonuc:ear Leukocyte Elastase Via Hydrolysis of N-t-Boc-alanyl-prolylalanine-p-nitroanilide (Boc-AAPAN) or N-t-Boc-alanyl-prolylvaline-pnitroanilide (Boc-AAPVN) Reagent:

0.05M TES (N-tris[hydroxymethyl]methyl-2-amino-ethanesulfonic acid) Buffer. pH 7.5.

0.2 mM Boc-AAPAN or Boc-AAPVN.

To prepare substrate, the solid was first dissolved in 10.0 ml DMSO. Buffer at pH 7.5 was then added to a final volume of 100 ml.

Crude extract of human polymorphonuclear leukocytes (PMN) containing elastase activity. Inhibitors (azetidinones) to be tested dissolved in DMSO just before use.

Assay Procedure:

50

To 1.0 ml of 0.2 mM Boc-AAPAN in a cuvette. 0.01-0.1 ml of DMSO with or without inhibitor was added. After mixing, a measurement was taken at 410 mμ to detect any spontaneous hydrolysis due to presence of test compound. 0.05 Milliliters of PMN extract was then added and the ΔOD/min at 410 mμ was measured and recorded. Beckman model 35 spectrophotometer was used.

Results:

Results in Table I were reported as ID50, effective dosage in micrograms per milliliter (µg/ml) for 50% inhibition of the enzyme activity 2 minutes after zero time.

Results were also expressed as Ki, the micromolar concentration of the inhibitor (μ M) giving 50% of the control enzyme activity; or as *obs/l which is the second order rate constant in per mole per second for inactivation of the enzyme.

O Comments:

The elastase activity in the crude PMN extract may vary from one preparation to another. A control of each new batch is run, and the volume added in the assay procedure is adjusted according to activity.

Accordingly, the compounds of Formula (I) can be used to reduce inflammation and relieve pain in diseases such as emphysema, rheumatoid arthritis, osteoarthritis, gout, bronchial inflammation, atherosclerosis, sepsis, septicemia, shock, periodontitis, cystic fibrosis, infectious arthritis, rheumatic fever and the like.

For treatment of inflammation, fever or pain, the compounds of Formula (I) may be administered orally, topically, parenterally, by inhalation spray or rectally in dosage unit formulations containing conventional non-toxic pharmaceutically acceptable carriers, adjuvants and vehicles. The term parenteral as used herein includes subcutaneous injections, intravenous, intramuscular, intrasternal injection or infusion techniques. In addition to the treatment of warm-blooded animals such as mice, rats, horses, dogs, cats, etc., the compounds of the invention are effective in the treatment of humans.

use, for example, as tablets, troches, lozenges, aqueous or oily suspensions, dispersible powders or granules, emulsions, hard or soft capsules, or syrups or elixirs. Compositions intended for oral use may be prepared according to any method known to the art for the manufacture of pharmaceutical compositions and such compositions may contain one or more agents selected from the group consisting of sweetening agents, flavoring agents, coloring agents and preserving agents in order to provide pharmaceutically elegant and palatable preparation. Tablets contain the active ingredient in admixture with non-toxic pharmaceutically acceptable excipients which are suitable for the manufacture of tablets. These excipients may be for example, inert diluents, such as calcium carbonate, sodium carbonate, lactose, calcium phosphate or sodium phosphate: granulating and disintegrating agents, for example, corn starch, or alginic acid: binding agents, for example starch, gelatin or acacia, and lubricating agents, for example magnesium stearate, stearic acid or talc. The tablets may be uncoated or they may be coated by known techniques to delay disintegration and absorption in the gastrointestinal tract and thereby provide a sustained action over a longer period. For example, a time delay material such as glyceryl monostearate or giyceryl distearate may be employed.

Formulations for oral use may also be presented as hard gelatin capsules wherein the active ingredient is mixed with an inert solid diluent, for example, calcium carbonate, calcium phosphate or kaolin, or as soft gelatin capsules wherein the active ingredient is mixed with water or an oil medium, for example peanut oil, liquid paraffin, or olive oil.

Aqueous suspensions contain the active materials in admixture with excipients suitable for the manufacture of aqueous suspensions. Such excipients are suspending agents, for example sodium carboxymethylcellulose, methylcellulose, hydroxypropylmethylcellulose, sodium alginate, polyvinylpyrrolidone, gum tragacanth and gum acacia; dispersing or wetting agents may be a naturally-occurring phosphatide, for example lecithin, or condensation products of an alkylene oxide with fatty acids, for example polyoxyethylene stearate, or condensation products of ethylene oxide with long chain aliphatic alcohols, for example heptadeca-ethyleneoxycetanol, or condensation products of ethylene oxide with partial esters derived from fatty acids and a hexitol such as polyoxyethylene sorbitol monooleate, or condensation products of ethylene oxide with partial esters derived from fatty acids and hexitol anhydrides, for example polyoxyethylene sorbitan monooleate. The said aqueous suspensions may also contain one or more preservatives, for example ethyl, or n-propyl, p-hydroxybenzoate, one or more coloring agents, one or more flavoring agents, and one or more sweetening agents, such as sucrose or saccharin.

Oily suspension may be formulated by suspending the active ingredient in a vegetable oil, for example arachis oil, olive oil, sesame oil or coconut oil, or in a mineral oil such as liquid paraffin. The oily suspensions may contain a thickening agent, for example beeswax, hard paraffin or cetyl alcohol. Sweetening agents such as those set forth above, and flavoring agents may be added to provide a palatable

oral preparation. These compositions may be preserved by the addition of an antioxidant such as ascorbic acid.

Dispersible powders and granules suitable for preparation of an aqueous suspension by the addition of water provide the active ingredient in admixture with a dispersing or wetting agent, suspending agent and one or more preservatives. Suitable dispersing or wetting agents and suspending agents are exemplified by those already mentioned above. Additional excipients, for example sweetening, flavoring and coloring agents, may also be present.

The pharmaceutical compositions of the invention may also be in the form of oil-in-water emulsions. The oily phase may be a vegetable oil, for example olive oil or arachis oils, or a mineral oil, for example liquid paraffin or mixtures of these. Suitable emulsifying agents may be naturally-occurring gums, for example gum acacia or gum tragacanth, naturally-occurring phosphatides, for example soy bean, lecithin, and esters or partial esters derived from fatty acids and hexitol anhydrides, for example sorbitan monooleate, and condensation products of the said partial esters with ethylene oxide, for example polyoxyethylene sorbitan monooleate. The emulsions may also contain sweetening and flavoring agents.

Syrups and elixirs may be formulated with sweetening agents, for example glycerol, propylene glycol, sorbitol or sucrose. Such formulations may also contain a demulcent, a preservative and flavoring and coloring agents. The pharmaceutical compositions may be in the form of a sterile injectable aqueous or oleagenous suspension. This suspension may be formulated according to the known art using those suitable dispersing or wetting agents and suspending agents which have been mentioned above. The sterile injectable preparation may also be a sterile injectable solution or suspension in a non-toxic parenterally-acceptable diluent or solvent, for example as a solution in 1,3-butane diol. Among the acceptable vehicles and solvents that may be employed are water, Ringer's solution and isotonic sodium chloride solution. In addition, sterile, fixed oils are conventionally employed as a solvent or suspending medium. For this purpose any bland fixed oil may be employed including synthetic mono- or diglycerides. In addition, fatty acids such as oleic acid find use in the preparation of injectables.

The compounds of Formula (I) may also be administered in the form of suppositories for rectal administration of the drug. These compositions can be prepared by mixing the drug with a suitable non-irritating excipient which is solid at ordinary temperatures but liquid at the rectal temperature and will therefore melt in the rectum to release the drug. Such materials are cocoa butter and polyethylene glycols.

For topical use, creams, ointments, jellies, solutions or suspensions, etc., containing the anti-inflammatory agents are employed.

The amount of active ingredient that may be combined with the carrier materials to produce a single dosage form will vary depending upon the host treated and the particular mode of administration. For example, a formulation intended for the oral administration of humans may contain from 5 mg to 5 gm of active agent compounded with an appropriate and convenient amount of carrier material which may vary from about 5 to about 95 percent of the total composition. Dosage unit forms will generally contain between from about 25 mg to about 500 mg of active ingredient.

It will be understood, however, that the specific dose level for any particular patient will depend upon a variety of factors including the activity of the specific compound employed, the age, body weight, general health, sex. diet, time of administration, route of administration, rate of excretion, drug combination and the severity of the particular disease undergoing therapy.

EXAMPLE 1

1-p-nitrophenylsulfonyl-4-benzyloxycarbonyl azetidin-2-one

45

50

Diazabicycloundecane (152 mg, 1 mM) was added to a mixture of 205 mg (1 mM) 4-benzyloxycarbonyl azetidin-2-one and 181 mg (1 mM) p-nitrobenzenesulfonyl chloride in 10 ml methylene chloride at room temperature. After stirring 2-1/2 hours, the orange solution was washed with water, dried over MgSO₄, and concentrated in vacuo. The residue was chromatographed on silica gel in hexane/ethyl acetate to yield 64 mg (17%) of 1-p-nitrophenylsulfonyl-4-benzyloxycarbonyl azetidin-2-one.

NMR (CDCl₃): § 3.3 (2H, doublet-quartet), 4.8 (qt. 1H), 5.2 (s, 2H), 7.2 (s, 5H), 8.2 (mit. 4H).

EXAMPLE 2

1-Acetyl-3.3-dimethyl-4-acetoxyazetidin-2-one

Step A: Preparation of 2-methyl-prop-1-enylacetate

A mixture of 72 g (1 M) isobutyraldehyde. 153 g (1.5 M) acetic annydride and 12 g (0.125 M) potassium acetate was refluxed seven hours. The cooled reaction mixture was washed with water and stirred with 300 ml saturated NaHCO₃ at 0°C for 45 minutes. The organic phase was dried over K₂CO₃ to yield a yellow oil which was distilled at atmospheric pressure to give 35.41 g (31%) of 2-methyl-prop-1-enylacetate, b.p. 122-126°.

NMR (CDCl₃): δ 1.6 (s, 6H). 2.1 (s, 3H), 6.9 (mlt. 1H).

15

30

35

40

55

Step B: Preparation of 3-3-dimethyl-4-acetoxyazetidin-2-one

Chlorosulfonyl isocyanate (16 ml) was added to a solution of 22.8 g (0.2 M) 2-methyl prop-1-enyl acetate in 50 ml methylene chloride at 0° under nitrogen. After stirring at 0° for 20 hours, the reaction mixture was added to a mixture of 20 ml water, 90 g ice, 48 g NaHCO₃ and 16.6 g Na₂SO₃ and stirred at 0° for 30 minutes. This was then extracted with 300 ml CH₂Cl₂ and the organic phase washed with brine, dried over MgSO₄ and concentrated in vacuo to give 27.75 g oil which was chromatographed on silica gel in hexane/ethyl acetate to yield 2.17 g (8.5%) of 3.3-dimethyl-4-acetoxyazetidin-2-one.

NMR (CDCl₃): δ 1.2 (s, 3H), 1.3 (s. 3H), 2.2 (s, 3H), 5.6 (s, 1H).

Step C: Preparation of 1-acetyl-3.3-dimethyl-4-acetoxyazetidin-2-one

A mixture of 283.3 mg (1.8 mM) 3.3-dimethyl-4-acetoxyazetidin-2-one, 2 ml pyridine and 2 ml acetic anhydride was heated to 100° in a sealed tube for 36 hours. The reaction mixture was concentrated in vacuo and the residue chromatographed on silica gel in hexane ethyl acetate to yield 295 mg (82%) of 1-acetyl-3.3-dimethyl-4-acetoxyazetidin-2-one.

NMR (CDCl₃): δ 1.2 (s. 3H), 22 (s. 3H), 2.5 (s. 3H), 6.1 (s. 1H).

EXAMPLE 3

1-Acetyl-4-acetoxy-3-n-propylazitidin-2-one

Step A: Preparation of Pent-1-enyl acetate

A mixture of 86 g (1M) valeraldehyde. 153 g (1.5 M) acetic anhydride, and 12 g (0.125 M) potassium acetate, was refluxed for 8 hours. The cooled mixture was then stirred with 100 ml saturated aqueous NaHCO₃ for one hour. The organic phase is separated, dried over K₂CO₃, and distilled at 40 mm to yield 46.15 g (45%) of pent-1-enylacetate, b.p. 89 °C.

NMR (CDCl₃): δ 1.0 (tr, 3H), 1.2-2.0 (mlt., 4H), 2.1 (s, 3H), 4.7-5.6 (mlt. 1H), 7.0-7.3 (mlt., 1H).

Step B: Preparation of 4-acetoxy-3-n-propylazetidin-2-one

Eight hundred microliters of chlorosulfonyl isocyanate was added to a solution of 1.28 g (10 mM) pent-1-enyl acetate in 5 ml methylene chloride at 0° under nitrogen. After stirring at 0° 5 days, the reaction mixture was added dropwise to a mixture of 5 g ice, 1.15 ml water, 2.82 g NaHCO₃ and 1.0 g Na₂SO₃ and stirred at 0° for 30 minutes. The mixture was extracted with 2 X 25 ml methylene choride and the

EP 0 337 549 A1

combined organic phases washed with brine, dried over MgSO₄, and concentrated in vacuo. The residue was chromatographed on silica gel in hexane/ethyl acetate to yield 60 mg trans 4-acetoxy-3-n-propylazetidin-2-one (3.4%).

NMR (CDCl₃): § 1.0 (mlt., 3H), 1.7 (mlt., 4H), 2.2 (s, 3H), 3.2 (tr, 1H), 5.6 (s, 1H), 6.7 (lrs, 1H),

5

Step C: Preparation of 1-acetyl-4-acetoxy-3-n-propylazeticin-2-one

A mixture of 56 mg (0.33 mM) 4-acetoxy-3-propylazetidin-2-one. 1 ml acetic anhydride and 1 ml pyridine was stirred at 100° in a sealed tube for 24 hours. After concentrating in vacuo the residue was chromatographed on silica gel in hexane/ethyl acetate, to yield 16 mg (23%) 1-acetyl-4-acetoxy-3-n-propylazetidine-2-one.

NMR (CDCl₃): 8 1.0 (br tr, 3H), 1.7 (mlt., 4H), 2.2 (s, 3H), 2.4 (s, 3H), 3.2 (tr, 1H), 6.1 (d, 1H).

15

EXAMPLE 4

1-Acetyl-4-methylsulfonylazetidin-2-one

Step A: Preparation of 1-acetyl-4-methylthioazetidin-2-one

A mixture of 300 mg (2.6 mM) 4-methylthioazetidin-2-one, 10 ml acetic anhydride and 10 ml pyridine was stirred at 100° in a sealed tube 24 hours. After concentrating in vacuo, the residue was chromatographed on silica gel in hexane/ethyl acetate to yield 324 mg (78%) of 1-acetyl-4-methylthioazetidine-2-one. NMR (CDCl₃): § 2.4 (s. 3H), 2.41 (s. 3H), 3.2 (doublet-quartet, 2H), 5.1 (coublet-doublet, 1H).

30

25

Step B: Preparation of N-acetyl-4-methylsulfinylazetidin-2-cne

A mixture of 130 mg (0.82 mM) N-acetyl-4-methylmicazeticinone and 200 mg (0.93 nM) 80% m-chloroperbenzoic acid in 5 ml methylene chloride was stirred at room temperature 5 minutes. After removing the solvent in vacuo. The residue was chromatographed on 2-2000 μ - silica gel plates in hexane:ethyl acetate to yield 57 mg (40%) of 1-acetyl-4-methylsuffinylazetidine-2-one. NMR (CDCl₃): δ 2.4 (s. 3H), 2.6 (s. 3H), 3.5 (mlt., 2H), 4.9 (mit., 1H).

40

EXAMPLE 5

3-Azido-4-carboethoxy-1-(p-methoxyphenyl)azetidin-2-one

45

To a solution of 3.06 g of azidoacetyl chloride in 50 mi of CH_2Cl_2 was added dropwise a solution of 3.57 ml of triethylamine and 5.3 g of the imine formed from ethylgiyoxalate and p-anisidine in 50 ml CH_2Cl_2 , with cooling at such a rate that the reaction temperature remained below 5°. The reaction was then stirred at room temperature for three hours and then wasned sequentially with 1N HCl, saturated aqueous sodium bicarbonate, and saturated aqueous sodium chloride. The organic phase was dried over magnesium sulfate, filtered, and evaporated, and the crude residue was recrystallized from carbon tetrachloride hexane to afford 3.7 g, of 3-azido-4-carboethoxy-1-(p-methoxyphenyl)azetidine-2-one; m.p. 80-85°.

NMR (CDCl₃): § 7.2 (d, J=9, 2H), 6.75 (d, J=9, 2H), 4.9 (d, J=6, 1H), 4.6 (d, J=6, 1H), 4.25 (n, J=8, 2H)

NMR (CDCl₃): δ 7.2 (d. J=9, 2H), 6.75 (d. J=9, 2H), 4.9 (d. J=6, 1H), 4.6 (d. J=6, 1H), 4.25 (q. J=8, 2H), 3.7 (s. 3H), 1.25 (t. J=8, 3H).

55

EXAMPLE 6

EP 0 337 549 A1

4-Carboethoxy-3-chloro-1-(p-methoxyphenyl)azetidine-2-one

4-carboethoxy-3-chloro-1-(p-methoxyphenyl)azetidine-2-one was prepared by following the same procedure as described in Example 5 but using chloroacetyl chloride and the imine formed from ethylglyoxalate and p-anisidine as the starting material. The crude product was recrystallized from ether (hexane) to give 3.1 g of 4-carboethoxy-3-chloro-1-(p-methoxyphenyl)azetidine-2-one, m.p. 99-100°.

NMR (CDCl₃): δ 7.2 (d. J = 9, 2H), 6.8 (d. J = 9, 2H), 5.1 (d. J = 6, 1H), 4.7 (d. J = 6, 1H), 4.25 (q. J = 7, 2H), 3.7 (s. 3H), 1.25 (t. J = 7, 3H).

10

EXAMPLE 7

4-Carboethoxy-3-methoxy-1-(p-methoxyphenyl)azetidine-2-one

4-Carboethoxy-3-methoxy-1-(p-methoxyphenyl)azetidine-2-one was prepared by following the same procedure as described in Example 5 but using methoxyacetyl chloride as the starting material. After chromatography the compound crystallized as a white solid; m.p. 116-118°.

NMR (CDCl₃): δ 7.2 (d. J=9, 2H), 6.75 (d. J=9, 2H), 4.7 (d. J=5, 1H), 4.6 (d. J=5, 1H), 4.2 (q. J=5, 2H).

3.7 (s. 3H), 3.5 (s. 3H), 1.2 (t. J = 5, 3H).

EXAMPLE 8

25

. 30

4-Carboethoxy-1-(p-methoxyphenyl)-3-phenylazetidin-2-one

To a solution of 17 ml of triethylamine and 5.0 g of the imine formed from ethyl glyoxalate and panisidine in 100 ml of refluxing 1,2-dichloroethane was added dropwise over 2 hours a solution of 16 ml of freshly distilled phenylacetyl chloride in 50 ml of dichloroethane. After refluxing for three hours the reaction was worked-up as per the 3-azidoazetidinone. The crude residue was chromatographed to yield the cis and trans isomers of 4-carboethoxy-1-(p-methoxyphenyl)-3-phenylazetidin-2-one as oils; cis: NMR (CDCl₃): δ 7.2 (m. 7H), 6.7 (d. J=9, 2H), 4.7 (s. 2H), 3.6 (s. 3H), 3.6 (q. J=7, 2H), 0.7 (t. J=7, 3H); trans: NMR (CDCl₃): 3 (m. 7H), 6.8 (d. J=9, 2H), 4.5 (d. J=2, 1H), 4.45 (d. J=2, 1H), 4.1 (q. J=7, 2H), 3.6 (s. 3H), 1.2 (t. J=7, 3H).

ەد

EXAMPLE 9

4-Carboethoxy-1-(p-methoxyphenyl)-3-vinylazetidin-2-one

45

4-Carboethoxy-1-(p-methoxyphenyl)-3-vinylazetidine-2-one was prepared by following the same procedure as described in Example 8 but using crotonyl chloride as the reagent. After chromatography the cis and trans isomers of the compound were obtained; cis (m.p. $70-72^{\circ}$), NMR (CDCl₃): δ 7.2 (d. J=9, 2H), 6.8 (d. J=9, 2H), 5.2-5.8 (m. 3H), 4.6 (d. J=6, 1H), 4.2 (m. 3H), 3.7 (s. 3H), 1.2 (t. J=7, 3H); trans (oii), NMR (CDCl₃): δ 7.25 (d. J=9, 2H), 6.8 (d. J=9, 2H), 5.7-6.2 (m. 1H), 5.2-5.5 (m. 2H), 4.25 (br.s., $\overline{1H}$), 4.2 (q. J=7, 2H), 3.9 (dd. J=1, Jz=6, 1H), 3.75 (s. 1H), 1.25 (t. J=7, 3H).

EXAMPLE 10

55

4-Carboethoxy-3-ethyl-1-(p-methoxyphenyl)azetidin-2-one

The cis and trans isomers of 4-carboethoxy-3-vinyl-1-(p-methoxyphenyl)azetidine-2-one are each hydrogenated with palladium on carbon in ethanol to yield the corresponding cis and trans isomers of 4-carboethoxy-3-ethyl-1-(p-methoxy-phenyl)azetidine-2-one.

5

EXAMPLE 11

4-Carboethoxy-1-(p-methoxyphenyl)-3-(N-methyl-trifluoroacetamido)azetidin-2-one

A solution of 2.16 g of 3-azido-4-carboethoxy-1-(p-methoxyphenyl)-azetidine-2-one in ethanol was hydrogenated with palladium to yield 4-carboethoxy-1-(p-methoxyphenyl)-3-aminoazetidin-2-one. This amine was acylated with 1.1 ml of trifluoro acetic anhydride in 10 ml CH₂Cl₂ containing 1.5 ml pyridine. followed by methylation using 1 ml dimethyl sulfate in 30 ml acetone containing 3 g potassium carbonate. After isolation, the crude product was crystallized to give 2.2 g of 4-carboethoxy-1-(p-methoxyphenyl)-3-(N-methyltrifluoroacetamido)azetidine-2-one, m.p. 102-104°.

NMR (CDCl₃): δ 7.2 (d, J=9, 2H), 6.75 (d, J=9, 2H), 5.5 (d, J=6, 1H), 4.7 (d, J=6, 1H), 4.2 (q, J=7, 2H), 3.7 (s, 3H), 3.2 (br.s., 3H), 1.2 (t, J=7, 3H).

20

EXAMPLE 12

25

45

4-Carboethoxy-3-methoxyazetidin-2-one

To a solution of 1.4 g of 4-carboethoxy-3-methoxy-1-(p-methoxyphenyl)azetidine-2-one in 50 ml acetonitrile at 0 was added a solution of 8.23 g of cerric ammonium nitrate in 50 ml H_2O over 3 minutes. After stirring at 0 for 1 hour the solution was poured into 200 ml of 10% sodium sulfite and extracted with 3 X 75 ml of ethyl acetate. The combined organic extracts were washed with 10% sodium sulfite and saturated sodium chloride solutions and dried over sodium sulfate. Filtration and evaporation gave an amber oil which was recrystallized from methylene chloride hexane to give 700 mg of 4-carboethoxy-3-methoxyazetidine-2-one; m.p. 91-92.

NMR (CDCl₃): δ 7.1 (br.s. 1H), 4.7 (dd. $J_1 = 2$, $J_2 = 5$, 1H), 4.3 (d, J = 5, 1H), 4.15 (q, J = 7, 2H), 3.4 (s, 3H). 1.25 (t, J = 7, 3H).

Following substantially the same procedure as described in Example 12 but using an appropriate 3-substituted azetidinone compounds (a) - (f) were prepared:

- (a) 4-Carboethoxy-3-chloroazetidin-2-one
- 20 NMR (CDCl₃): δ 7.3 (br.s., 1H), 5.0 (dd, $J_4 = 2$, $J_2 = 6$, 1H), 4.4 (d, J = 6, 1H), 4.2 (q, J = 7, 2H), 1.3 (t, J = 7, 3H).
 - (b) 4-Carboethoxy-3-phenylazetidin-2-one-2-(cis and trans)

NMR (CDCl₃): cis: δ 7.2 (s, 5H), 6.4 (br.s., 1H), 4.7 (d, J=6, 1H), 4.4 (d, J=6, 1H), 3.7 (q, J=7, 2H), 0.75 (t, J=7, 3H); trans δ 7.2 (s, 5H), 6.9 (br.s., 1H), 4.3 (br.d., J=2, 1H), 4.1 (q, J=7, 2H), 4.0 (d, J=2, 1H), 1.2 (t, J=7, 3H).

- (c) 4-Carboethoxy-3-(N-methyltrifluoroacetamido) azetidin-2-one
- NMR (CDCl₃): δ 7.2 (br.s., 1H), 5.4 (d, J=6, 1H), 4.5 (d, J=6, 1H), 4.15 (q, J=7, 2H), 3.2 (s. 3H), 1.2 (t. J=7, 3H).
 - (d) 4-Carboethoxy-3-vinylazetidin-2-one(cis and trans)
- NMR (CDCl₃) cis: δ 7.1 (br.s., 1H), 5.2-5.8 (m, 3H), 4.0-4.4 (m, 4H), 1.25 (t, J = 7, 3H); trans: δ =7.25 (br.s., 1H), 5.0-6.2 (m, 3H), 4.1 (q, J = 7, 2H), 3.9 (d, J = 2, 1H), 3.7 (dd, J₁ = 2, J₂ = 7, 1H), 1.2 (t, J = 7, 3H).
 - (e) 4-Carboethoxy-3-ethylazetidin-2-one

Cis: NMR($\overline{\text{CDCI}_3}$): δ 6.9 (br. s., 1H); 4.2 (m, 3H); 3.4 (dd, $J_1 = 6$, $J_2 = 8$, 1H); 1.51 (q, J = 8, 2H); 1.2 (t, J = 7, 3H); 1.0 (t, J = 8, 3H).

55 Trans: NMR(CDCl₃): δ 6.8 (br. s., 1H); 4.2 (q, J=7, 2H); 3.8 (d, J=2, 1H); 3.2 (dd, J₁=2, J₂=7, 1H); 1.8 (- (dq, J₁=2, J₂=8, 2H); 1.2 (t, J=7, 3H); 1.0 (t, J=8, 3H).

(f) 3-Azido-4-carboethoxyazetidin-2-one

EXAMPLE 13

4-Carboethoxy-3-(N-methyltrifluoroacetamido)azetidine-2-one-1-sulfonic acid tetrabutylammonium salt

To a solution of 140 mg of 4-carboethoxy-3-(N-methyltrifluoroacetamido)azetidine-2-one in 5 ml of pyridine at 80° was added 250 mg of sulfur trioxide pyridine complex, and the resulting mixture was stirred for 30 minutes at 80°. The solution was poured into 100 ml of 0.5 N KH₂PO₄ and extracted with 2 X 25 ml of methylene chloride. The combined organic washes were back-extracted with 25 ml of KH₂PO₄ solution. The combined aqueous phases were then treated with 680 mg of tetrabutylammonium hydrogen sulfate and extracted with 3 X 50 ml of methylene chloride. After drying (sodium sulfate) and evaporation of the organic phase the crude 4-carboethoxy-3-(N-methyltrifluoroacetamido)azetidine-2-one-1-sulfonic acid tetrabutylammonium salt was chromatographed to yield an oil.

NMR (CDCl₃): δ 5.3 (d, J = 6, 1H), 4.7 (d, J = 6, 1H), 4.15 (q, J = 7, 2H), 3.2 (m, 11H), 0.8-1.8 (m, 31H).

Applying the same procedure as described above, the following tetrabutylammonium salts of other azetidine derivatives were prepared:

(a) 4-Carboethoxy-3-methoxyazetidin-2-one-1-sulfonic acid tetrabutylammonium salt NMR ($\overline{\text{CDCl}_3}$): δ 4.55 (d. J=6, 1H), 4.5 (d. J=6), 1H), 4.1 (q. J=7, 2H), 3.4 (s. 3H), 3.2 (m. 8H), 0.8-1.8 (m. 31H).

(b) 4-Carboethoxy-3-vinylazetidin-2-one-1-sulfonic acid tetrabutylammonium salt

EXAMPLE 14

30

5

10

20

25

4-Carboethoxy-1-(p-nitrobenzenesulfonyl)-3-phenylazetidin-2-one

To a solution of 720 mg of 4-carboethoxy-3-trans-phenylazetidin-2-one-in 20 ml methylene chlonde at 0° were added sequentially 595 mg of p-nitrobenzenesulfonyl chloride and 0.48 ml of DBU. The solution was stirred for several hours, diluted with 50 ml of methylene chloride, washed once with water and dried over sodium sulfate. Filtration and evaporation gave a crude residue which was chromatographed to yield pure 4-carboethoxy-1-(p-nitrobenzenesulfonyl)-3-phenyl-azetidin-2-one.

NMR (CDCl₃): δ 8.3 (d, J=9, 2H), 8.2 (d, J=9, 2H), 7.2 (br.s., 5H), 4.0 (q, J=7, 2H), 3.7 (m, 2H), 1.2 (t, L) J=7. 3H). Similarly prepared was the corresponding cis-3-phenyl compound. NMR (CDCl₃): δ 8.4 (d. J=9. 2H). 8.25 (d. J = 9. 2H). 7.2 (s. 5H). 5.0 (s. 1H). 3.7 (m. 3H). 0.85 (t. 5 = 7. 3H).

Following the same procedure as described above but using appropriate reagents, the following compounds were prepared:

(a) 4-Carboethoxy-1-(p-nitrobenzensulfonyl)-3-vinylazetidin-2-one NMR ($\overline{\text{CDCl}_1}$): cis: δ 8.3 (d. J = 9, 2H), 8.2 (d. J = 9, 2H), 5.2-6.0 (m. 3H), 4.0-4.6 (m. 4H), 1.2 (t. J = 7, 3H); trans: δ 8.2 (d. J=9, 2H), 8.15 (d. J=9, 2H), 5.2-6.0 (m, 3H), 3.9-4.4 (m, 4H), 1.25 (t. J=7, 3H).

- (b) 4-Carboethoxy-3-ethyl-1-(p-nitrobenzenesulfonyl)azetidin-2-one
- (c) 3-Azido-4-carboethoxy-1-(p-nitrobenzenesulfonyi) azetidin-2-one
- (d) 4-Carboethoxy-3-chloro-1-(p-nitrobenzensulfonyl)azetidin-2-one

EXAMPLE 15

55

45

50

4-Carboethoxy-3-phenyl-1-trifluoromethanesulfenylazetidin-2-one

EP 0 337 549 A1

To a mixture of 1.2 g of 4-carboethoxy-3-phenylazetidin-2-one and 1.2 ml of triethylamine in 25 ml of methylene chloride at 0 was added dropwise over 10 minutes 11.25 ml of a 10% solution of trifluoromethanesulfenyl chloride in ether. After stirring for several hours the solution was washed with water. dried over sodium sulfate, filtered and evaporated. The crude residue was chromatographed to yield pure 4carboethoxy-3-phenyl-1-trifluoromethanesulfenylazetidin-2-one as an oil. NMR (CDCl₃): δ 7.2 (s, 5H), 4.6 (d, J = 3, 1H), 4.3 (m, 3H), 1.3 (t, J = 7, 3H).

EXAMPLE 16

10

1-Tosyloxymethyl-3-n-Propyl-4-p-nitrophenylthioazetidin-2-one

15

Step A: Preparation of 3-Propyl-4-p-nitrophenylthio azetidin-2-one

3-Propyl-4-acetoxy azetidinone. 171 mg, is refluxed with 200 mg p-nitrophenyl thiol in 10 ml benzene for 6 hours. The solution is washed 3x with aqueeus Na₂CO₃, dried with MgSO₄, filtered and evaporated. The residue is chromatographed on silica gel, eluting with 10:1 CHCl3-EtOAc, affording 3-Propyl-4-pnitrophenylthioazetidin-2-one.

Step B: Preparation of 1-Tosyloxymethyl-3-n-propyl-4-p-nitrophenylthio azetidin-2-one

25

3-Propyl-4-p-nitrophenylthioazetidine-2-one, 266 mg, is stirred overnight at room temperature with 0.25 ml aqueous formalin (37%) and 17 mg K2CO3. Water and formaldehyde are removed in vacuo, and flushed with 2 ml pyridine. The residue is taken up in 4 ml pyridine and treated for 1 hour at room temperature with 200 mg p-toluenesulfonyl chloride. The pyridine is evaporated and replaced with 5 ml benzene. The solution is washed with aqueous H3PO4 and then aqueous K2HPO4, dried with MgSO4, filtered and evaporated. The residue is chromatographed on silica gel, eluting with 25:1 CHCl3-EtOAc, providing 1tosyloxymethyl-3-n-propyl-4-p-nitrophenylthio-azetidin-2-one.

35

EXAMPLE 17

40

1-Tosyloxymethyl-3-n-propyl-4-p-nitrophenylsulfinyl azetidin-2-one

45

1-Tosyloxymethyl-3-n-propyl-4-p-nitrophenylsulfinylazetidin-2-one. 450 mg, is treated for 1.2 hour in 10 ml CH2Cl2 with 172 mg m-chloroperbenzoic acid. The solution is washed with aqueous K2HPO4, dried with MgSO₄, filtered and evaporated, leaving pure 1-tosyloxymethyl-3-n-propyl-4-p-nitrophenylsulfinyl azetidine-2-one.

50

1-Acetoxymethyl-4-p-nitrophenylsulfinyl-3-n-propylazetidin-2-one

55

Step A: Preparation of 3-n-propyl-4-p-nitrophenylthioazetidin-2-one

3-n-Propyl-4-acetoxya zetidinone (1.164 g. 6.58 mmole) and 1.02 g (6.58 mmole) p-nitrothiophenol were heated in a tube in the steam bath for 3.5 hours. The reaction mixture was cooled, diluted with 100 ml ethyl acetate, and the organic phase was washed with 100 ml water, 70 ml 1M H₃PO₄ and 3x100 ml saturated

EP 0 337 549 A1

 K_2CO_3 . The organic phase was dried over magnesium sulfate, filtered, and solvent removed in vacuo to yield 1.53 g of yellow crystals which were chromatographed on a silica gel column in chloroform-ethyi acetate (4:1) to give 359 mg (19%) of 3-n-propyl-4-p-nitrophenylthioazetidin-2-one. NMR (CDCl₃): δ 0.92 (tr, 3H), 1.2-1.6 (br m, 4H), 3.10 (tr, 1H), 4.91 (d. 1H), 7.0 (br s, 1H), 7.50 (d, 2H), 8.20

(d, 2H).

Step BPreparation of 1-Acetoxymethyl-4-p-nitrophenylthio-3-n-propylazetidin-2-ope

A mixture of 273 mg (0.94 mmole) azetidinone from Step A, 26.3 mg paraformaldehyde and 178 mg (0.56 mmole) cesium carbonate was stirred in 20 ml dry tetrahydrofuran at ambient temperature 16.5 hours under nitrogen. A mixture of 430 µl pyridine and 2.56 ml acetic anhydride was added to the reaction mixture and the stirring continued 5 more hours. The solvents were removed in vacuo to give 604 mg crude product which was chromatographed on a silica gel flash column in hexane-ethyl acetate 3.1. This gave 102 mg (30%) of 1-acetoxymethyl-4-p-nitrophenylthio-3-n-propylazetidin-2-one.

NMR (CDCl₃): δ 1.0 (tr. 3H), 1.2-1.85 (br m, 4H), 2.1 (s, 3H), 3.22 (tr. 1H), 4.95 (d. 1H), 5.18 (ABBA pattern, $J_1 = 30H_3$, $J_2 = 5H_3$, 2H), 7.65 (d. 2H), 8.22 (d. 2H).

Step CPreparation of 1-Acetoxymethyl-4-p-nitrophenylsulfinyl-3-n-propylazetidin-2-one

To a solution of 46 mg (0.127 mmole) azetidinone from Step B in 4 ml CH₂Cl₂ and 4 ml saturated aqueous NaHCO₃ was added 27 mg (0.127 mM) 80% m-chloroperbenzoic acid and the reaction mixture stirred vigorously 15 minutes. The phases were separated and the organic phase was dried over MgSO₄. filtered and stripped to yield 57 mg crude product which was chromatographed on a 1000 μ silica gel prep TLC plate in chloroform-ethyl acetate 4:1 to yield 15 mg (31%) of 1-acetoxymethyl-4-p-nitrophenylsulfinyl-3-n-propylazetidin-2-one.

NMR (CDCl₃): δ 0.93 (tr. 3H), 1.2-1.8 (br m, 4H), 2.1 (s, 3H), 3.55 (tr. 1H), 4.66 (d. 1H), 5.04 (ABBA pattern,

 $J_1 = 34H_3$, $J_2 = 6H_3$, 2H), 8.2 (d. 2H), 8.52 (d. 2H).

30

EXAMPLE 19

35

4-Acetoxy-3-n-propylazetidin-2-one-1-sulfonic acid tetrabutylammonium salt

A solution of 82 mg (0.463 mmole) 3-propyl-4-acetoxy azetidin-2-one in 5 ml pyridine was heated to 80°. 221 Mg (1.39 mmole) sulfur trioxide-pyridine complex was added and the reaction mixture stirred at 80° one hour. It was then poured into 100 ml 0.5M KH₂PO₄ (aquecus) and washed with 2x25 ml CH₂Cl₂. The combined organic washes were backwashed with 25 ml 0.5M KH₂PO₄. 157 Mg (0.463 mmole) Bu₄NHSO₂ was added to the combined aqueous phases. This was extracted with 2x25 ml CH₂Cl₂ and the combined extracts were dried over MgSO₄, filtered, and stripped in vacuo to yield 12.4 mg of an oily residue which was chromatographed on a small silica gel column, eluted first with 75 ml hexane/ethyl acetate (3:1) to remove starting material, then with 100 ml ethyl acetate/methanol (4:1) to yield 13 mg (5.7%) 4-acetoxy-3-n-propylazetidin-2-one-1-sulfonic acid tetrabutylammonium salt.

NMR (CDCl₃): δ 1.0 (m, 16H), 1.75 (br m, 20H), 2.16 (s, 3H), 2.90 (br s, H), 3.1 (tr. 1H), 3.3 (tr. 8H), 4.08 (br tr. 1H), 6.18 (s, 1H).

50

EXAMPLE 20

55 (3R.4S)-1-(benzylaminocarbonyl)-3-ethyl-3-methyl-4-(4-carboxy)phenoxyazetidin-2-one

Step A: Preparation of (3R.4S)-1-t-butyldimethylsilyl-3-methylazetidin-2-one-4-carboxylic acid

To a solution of 27.5 ml of diisopropylamine in 150 ml of THF at -20°C was added 73.5 ml of 2.4N n-butyl lithium in hexane. After 15 minutes, the solution was cooled to -70°C and a solution of 20 gm of (4S)-1-t-butyldimethylsilylazetidin-2-one-4-carboxylic acid in 75 mL of THF was added. The solution was warmed to -20°C for 15 minutes before a solution of 13.5 mL of methyl iodide in 20 mL of THF was added. After 30 minutes at -20 to 0°C, the reaction was diluted with 300 mL of ether and then poured into a mixture of ice and 400 mL of 1N HCl. The layers were separated and the aqueous layer extracted with ether. The ether layers were washed with brine, dried over sodium sulfate and evaporated. The residue was crystallized from hexane to give 12-15 gms of (3R,4S)-1-t-butyldimethylsilyl-3-methylazetidin-2-one-4-carboxylic acid. NMR (CDCl₃): δ .14 (2, 3H), .32 (s, 3H), .91 (d, 3H), .98 (s, 9H), 3.34 (dq, 1H), 3.71_(d, 1H)

10

Step B: Preparation of (3R,4S)-1-t-butyldimethylsilyl-3-ethyl-3-methylazetidin-2-one-4-carboxylic acid

To a solution of 13 mL of diisopropylamine in 75 mL of THF at -20° C was added 35 mL of 2.4 M n-butyl lithium in hexane. After 15 minutes the solution was cooled to -70° C and a solution of 10 gms of (3R.4S)-1-t-butyldimethylsilyl-3-methylazetidin-2-one-4-carboxylic acid in 50 mL of THF was added. The solution was warmed to -20° C for 15 minutes and a solution of 6.7 mL of ethyl iodide in 10 mL of THF was added. After 30 minutes at -20° to 0° C the reaction was diluted with ether and poured into a mixture of ice and 1 N HCl. The layers were separated and the aqueous layer extracted with ether. The ether layers were each washed with brine, dried over sodium sulfate and evaporated. The residue was crystallized from a minimum amount of hexane to give 8.8 gms of (3R.4S)-1-t-butyldimethylsilyl-3-ethyl-3-methylazetidin-2-one-4-carboxylic acid.

NMR(CDCl₃): δ .15 (s, 3H), .31 (s, 3H), .98 (s, 9H), 1.04 (t, 3H), 1.22 (s, 3H), 1.78 (q, 2H), 3.94 (s. 1H).

25

Step C: Preparation of (3R, 4S)-3-ethyl-3-methyl-4-(4-carbo-t-butoxy)phenoxyazetidin-2-one

To a solution of 13.0 gms of (3R. 4S)-1-t-butyldimethylsilyl-3-ethyl-3-methylazetidin-2-one-4-carboxylic acid in 75 mL of DMF and 15 mL of acetic acid under N₂ was added 23 gms of lead tetraacetate. The reaction was heated at 45-50°C for 18 hours and then poured into ice water and extracted into 2 portions of ether. The ether layers were washed with water, dilute sodium bicarbonate solution and brine, dried over sodium sulfate and evaporated to give 13 gm of crude oil containing a mixture of (3R, 4S) and (3R, 4R)-4-acetoxy-3-ethyl-3-methylazetidin-2-one. To this mixture in 50 mL of acetone was slowly added a solution of 14 gms of t-butyl 4-hydroxybenzoate in 50 mL of acetone, 5 mL of water and 29 mL of 2N sodium hydroxide. The reaction was stirred at room temperature for 64 hours and then diluted with water and extracted with 2 portions of ether. The ether layers were washed with brine, dried over sodium sulfate and evaporated. The residue was prep LC'ed with 15-25% ethylacetate-hexanes to give 6.3 gm of the higher R, (4R) ether and 1.5 gm of the desired (3R, 4S)-3-ethyl-3-methyl-4-(4-carbo-t-butoxy)phenoxyazetidin-2-one. NMR (CDCl₃): δ 1.0 (t, 3H), 1.38 (s, 3H), 1.54 (s, 9H), 1.6-2.0 (m, 2H), 5.30 (s, 1H) 6.7 (brs. 1H), 6.78 (d, 2H), 7.90 (d, 2H).

Step D: Preparation of discrete description of phenoxyazetidin-2-one of discrete description description of discrete description of description of description of description of description of descri

45

To a solution of 1.5 gm of (3R, 4S)-3-ethyl-3-methyl-4-(4-carbo-t-butoxy)phenoxyazetidin-2-one in 25 mL of methylene chloride was added 1.2 mL of benzyl isocyanate, 1.4 mL of triethylamine and 10 mg of 4-dimethylaminopyridine. The reaction was stirred at room temperature for 16 hours and then evaporated. The residue was flash chromatographed eluting with 10 to 25% EtoAc Hexane to give 2.3 gm of (3R, 4S)-1-(benzylaminocarbonyl)-3-ethyl-3-methyl-4-(4-carbo-t-butoxy)phenoxy azetidin-2-one. NMR (CDCl₃): δ .98 (t. 3H), 1.36 (s. 3H) 1.50 (s. 9H), 1.62 (m. 1H), 1.84 (m. 1H), 4.42 (d. 2H), 5.64 (s. 1H), 6.80 (brt, 1H), 7.06 (d. 2H), 7.24 (brs, 5H), 7.90 (d. 2H).

Step E: Preparation of (3R, 4S)-1-(benzylaminocarbonyl)-3-ethyl-3-methyl-4-(4-carboxy)phenoxyazetidin-2-one

To 2.3 gms of (3R, 4S)-1-(benzylaminocarbonyl)-3-ethyl-3-methyl-4-(4-carbo-t-butoxy) phenoxyazetidin-

EP 0 337 549 A1

2-one in an ice bath under N₂ was added 5 mL of anisole and then 25 mL of precooled trifluoroacetic acid. After 1.5 hours at 0 °C, the volatiles were removed in vacuo without heating and the residue flash chromatographed using hexane, then 15% EtoAc.Hexane, then 1° HoAc in 15% EtoAc.hexanes to give after ether trituration 1.8 gms of (3R. 4S)-1-(benzylaminocarbonyl)-3-ethyl-3-methyl-4-(4-carboxyl-phenoxyazetidin-2-one

NMR (CDCl₃): § 1.03 (t, 3H), 1.46 (s, 3H), 1.66 (m, 1H), 1.94 (m, 1H), 4.50 (d, 2H), 5.76 (s, 1H), 6.9 (brt, 1H), 7.05 (d, 2H), 7.25 (brs, 5H), 7.98 (d, 2H).

EXAMPLE 21

Starting with 3,3-diethyl-4-acetoxyazetidin-2-one as prepared in Scheme (d) followed by displacement of the acetate with the appropriate phenol and acylation of the nitrogen with the corresponding chiral isocyanate as shown in Scheme (h) and example 20, steps C-E, the following compounds were prepared. The diastereomers obtained on acylation were separated by silica gel chromatography using 10-30% ethylacetate-hexane solvent mixtures.

 $20 \hspace{0.1in} \textbf{(4S)-3.3-diethyl-1-((R)-\alpha-ethylbenzylaminocarbonyl)-4-(4-carboxymethyl)} phenoxyazetidin-2-one.$

10

35

45

50

NMR (CDCl₃): δ 0.9 (t,3H,J = 7Hz), 0.94 (t,3H,J = 7Hz), 1.07 (t,3H,J = 7hz) 1.65 - 2.05 (m,6H), 3.58 (s,2H). 4.8 (q,1H, J = 8Hz), 5.58 (s,1H), 7.0 (d, 1H, J = 8Hz), 7.1 - 7.45 (m,9H)

25 (4S)-3.3-diethyl-1- $((R)-\alpha-n$ -propylbenzylaminocarbonyl)-4-(4-carboxymethyl)phenoxyazetidin-2-one.

NMR (CDCl₃): δ 0.91 (t,3H,J=7Hz), 0.94 (t,3H,J=7Hz), 1.07 (t,3H,J=7hz) 1.34 (m,2H), 1.65 -2.05 (m,6H), 3.57 (s,2H), 4.88 (q, 1H, J=7Hz), 5.58 (s,1H), 7.0 (d, 1H, J=7Hz) 7.1 - 7.5 (m, 9H)

30 (4S)-3.3-diethyl-1- $((R)-\alpha$ -allyl-(4-methyl)benzyiaminocarbonyl)-4-(4-carboxymethyl)phenoxyazetidin-2-one.

NMR (CDCl₃): δ 0.96 (t,3H,J=7Hz), 1.07 (t,3H,J=7Hz), 1.7 - 2.1 (m, 4H), 2.32 (s, 3H), 2.57 (t,2H, J=7Hz), 3.58 (s, 2H), 4.95 (q, 1H, J=7Hz), 5.14 (m, 2H), 5.58 (s, 1H), 5.66 (m, 1H), 7.03 (d, 1H, J=7Hz), 7.16 (s, 4H), 7.19 (s, 4H).

(4S)-3.3-diethyl-1-((R)- α -allyl(3,4-methylenedioxy)benzylaminocarbonyl)-4-(4-carboxymethyl)-phenoxyazetidin-2-one.

NMR (CDCl₃): δ 0.96 (t.3H.J=7Hz), 1.05 (t.3H.J=7Hz), 1.65 - 2.05 (m, 4H), 2.54 (t. 2H J=6Hz) 4.87 ((q. 1H. \pm 0 J=7Hz), 5.05 -5.2 (m, 2H), 5.58 (s. 1H), 5.66 (m. 1H), 5.94 (s. 2H), 6.76 (s. 3H), 6.98 (d. 1H. J=7Hz), 7.2 (m.4H)).

(4S)-3.3-diethyl-1- $((R)-\alpha-n$ -propyl(3.4-methylenedioxy)-benzylaminocarbonyl)-4-(4-carboxymethyl)phenoxy-azetidin-2-one.

NMR (CDCl₃): δ 0.9 (t,3H,J=7Hz), 0.94 (t,3H,J=7Hz), 1.06 (t, 3H J=7Hz), 1.3 (m, 2H), 1.65 -2.1 (m, 6H), 3.58(s, 2H), 4.76(q, 1H, J=7hz), 5.58(s, 1H), 5.92 (s,2H), 6.15 (s, 3H) 6.88 (d, 1H, J=7Hz), 7.2 (m, 4H).

 $(4S)-3.3-diethyl-1-((R)-\alpha-n-propyl(4-methyl)benzylaminocarbonyl)-4-(4-carboxy)phenoxyazetidin-2-ne.$

NMR (CDCl₃): 5.0.91 (t.3H.J = 7Hz). 0.98 (t.3H,J = 7Hz). 1.07 (t, 3H. J = 7Hz) 1.32 (m, 2H), 1.65 - 2.1 (m, 6H). 2.33(s, 3H), 4.83(q, 1H, J = 7hz), 5.71(s, 1H), 6.93 (d, 1H, J = 7Hz), 7.16 (s, 4H), 7.25 (d,2H,J = 8Hz). 8.04 (d, 2H, J = 8Hz).

55 (4S)-3.3-diethyl-1-((R)-α-n-propyl(4-methyl)benzylaminocarbonyl)-4-(4-carboxymethyl)phenoxyazetidin-2-one.

NMR (CDCl₃): δ 0.9 (t,3H,J=7Hz), 0.93 (t,3H,J=7Hz), 1.07 (t, 3H, J=7Hz) 1.28 (m, 2H), 1.7 -2.1 (m, 6H), 2.33(s, 2H), 3.6 (s,2H), 4.81 (q, 1H, J=7hz), 5.56 (s, 1H), 6.93 (d, 1H, J=7Hz), 7.15 (s, 4H), 7.2 (s, 4H).

Claims

1. A compound of the formula (A)

70 \mathbb{R}^{1} $\mathbb{R}^{$

wherein

R and R' independently are C_{1-6} alkyl or C_{1-6} alkoxy C_{1-6} alkyl;

M is

(1) hydrogen.

(2) C_{1-6} alkyl,

(3) C2-5 alkenyl, or

(4) C₁₋₅ alkoxy C₁₋₆ alkyl;

X₅ is

(1) hydrogen.

(2) C₁₋₆ aikyl.

(3) halo C₁₋₅ alkyl.

(4) C₂₋₅ alkenyl,

(5) C_{2-5} alkynyl.

(6) carboxy.

(7) carboxy-C₁₋₅ alkyl.

(8) carboxy-C--s alkylcarbonyl.

(9) carboxy-C--s alkylcarbonylamino.

(10) carboxy-C₂₋₅ alkenyl,

(11) hydroxy-C: - alkyl.

(12) C. - s alkylcarbonyl.

(13) C1-5 alkylcarbonylamino, or

(14) di-(C: -salkyl)amino-C: -salkyl; and

X₅ is

(1) hydrogen,

(2) C1-5 alkyl.

(3) halo

(4) carboxy.

(5) C_{1-5} alkoxy,

(6) phenyl,

(7) C.-s alkylcarbonyl.

(8) di-(C1-salkyl)amino.

(9) phenoxy.

(10) methylenedioxy.

(11) 2.3-furanyl, or

(12) 2.3-thienyl; or

a pharmaceutically acceptable salt thereof.

2. A compound of Claim 1 wherein:

R and R¹ independently are C_{1-6} alkyl;

and

X₅ is carboxy or carboxy-C₁₋₅ alkyl.

3. A compound of Claim 2 wherein:

M is C₁₋₃ alkyl or allyl; and

X₅ is hydrogen, C₁₋₅ alkyl, or 3,4-methylenedioxy or phenyl.

4. A compound of Claim 3 wherein:

R is ethyl; and

R' is methyl or ethyl.

5. A compound of Claim 4 wherein:

5 R and R1 are ethyl:

M is n-propyl:

X5 is 4-carboxymethyl; and

X₅ is 4-methyl.

6. A pharmaceutical composition for the inhibition of human leukocyte elastase which comprises a nontoxic therapeutically effective amount of a compound of Claim 1 and a pharmaceutically acceptable carrier.

7. A composition of Claim 6 wherein:

R is ethyl:

R1 is methyl or ethyl:

15 M is C.-3 alkyl or allyl;

X₅ is carboxy or carboxy-C₁₋₆ alkyl;

and

 X_5 is hydrogen C_{1-5} alkyl. 3.4-methylenedioxy or phenyl.

8. A composition of Claim 7 wherein:

20 R1 is ethyl:

M is n-propyl:

X₅ is 4-carboxymethyl; and

X₅ is 4-methyl.

9. A process for the preparation of the compounds of Claim 1 which comprises

25 (1) reacting a compound of the following formula (B)

30

with a compound of the formula (C)

35

- wherein X5 is
 - (1) hydrogen
 - (2) C -- alkyl.
 - (3) halo-C. -; alkyl,
 - (4) C₂₋₅ alkenyl.
 - (5) C2-4 alkynyl.
 - (6) C: -5 alkoxycarbonyl.
 - (7) C: s alkoxycarbonyl-C: s alkyl.
 - (8) C. alkoxycarbonyi-C. alkylcarbonyi.
 - (9) C₁₋₅ alkoxycarbonyl-C₁₋₅ alkylcarbonylamino.
 - (10) C₁₋₅ alkoxycarbonyl-C₂₋₅ alkenyl.
 - (11) hydroxyaikyl.
 - (12) C. alkylcarbonyl.
 - (13) C: alkylcarbonylamino, or
- (14) di-(C--, alkyl)amino-C1-, alkyl under basic conditions to afford a compound of the formula (D)

41

$$R$$
 NH
 X_2
 (D)

and (2) reacting compound (D) with a compound of the formula (E)

under basic conditions, and optionally converting X5 into X5, to yield the compound of the formula (A)

$$\begin{array}{c|c}
R & & \\
\hline
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 &$$

10. A process of Claim 9 wherein:

Step (1) is in the presence of an alkali metal hydroxide. Ste (2) is in the presence of a $tri(C_{1-5} \text{ alkyl})$ -amine; and the conversion of X_5^+ into X_5 is accomplished in the presence of a strong acid. Claims for the following Contracting States: ES and GR

1. A process for the preparation of a compound of the formula (A)

$$\begin{array}{c|c}
R & X_5 \\
\hline
CONHCH & X_5
\end{array}$$
(A)

wherein

R and R¹ independently are C_{1-5} alkyl or C_{1-5} alkoxy C_{1-5} alkyl;

M is

5

20

30

35

40

45 (1) hydrogen.

(2) C. -5 alkyl.

(3) C_{2-5} alkenyl, or

(4) C--5 alkoxy-C--5 alkyl;

X₅ is

50 (1) hydrogen.

(2) C - - 5 alkyl.

(3) halo C: -; alkyl,

(4) C_{2-5} alkenyl,

(5) C₂₋₅ alkynyl.

55 (6) carboxy.

(7) carboxy-C₁₋₅ alkyl.

(8) carboxy-C1-5 alkylcarbonyl.

(9) carboxy-C.- alkylcarbonylamino.

- (10) carboxy-C2-5 alkenyl.
- (11) hydroxy-C: -5 alkyl.
- (12) C:-; alkylcarbonyl.
- (13) C₁₋₅ alkylcarbonylamino, or
- 5 (14) di-(C₁₋₅alkyl)amino-C₁₋₅alkyl; and

X₆ is

- (1) hydrogen,
- (2) C · 5 alkyl.
- (3) halo
- 10 (4) carboxy,
 - (5) C1-5 alkoxy.
 - (6) phenyl.
 - (7) C: -s alkylcarbonyl.
 - (8) di-(C:-6 aikyl)amino.
- 15 (9) phenoxy.
 - (10) methylenedioxy,
 - (11) 2.3-furanyl, or
 - (12) 2.3-thienyl; or
 - a pharmaceutically acceptable salt thereof which comprises
- 20 (1) reacting a compound of the following formula (B)

25

with a compound of the formula (C)

30

- wherein X₅ is
 - (1) hydrogen
 - (2) C - 5 alkyl.
 - (3) halo-C:-s alkyl,
 - (4) Cz-s alkenyl.
- 10 (5) C₂₋₅ alkynyl,
 - (6) C--; alkoxycarbonyl.
 - (7) C.-s alkoxycarbonyl-C:-s alkyl.
 - (8) C. -5 alkoxycarbonyl-C: -5 alkylcarbonyl.
 - (9) C: -5 alkoxycarbonyl-C1+5 alkylcarbonylamino.
- 5 (10) C₁₋₅ alkoxycarbonyl-C₂₋₅ alkenyl.
 - (11) hydroxyalkyl.
 - (12) C.-s aikylcarbonyl.
 - (13) C. -s alkylcarbonylamino, or
- (14) di-(C: +; alkyl)amino-C: +; aikyl
- under basic conditions to afford a compound of the formula (D)

$$R$$
 NH
 X_5
 (D)

55

and (2) reacting compound (D) with a compound of the formula (E)

under basic conditions, and optionally converting X_5 into X_5 , to yield the compound of the formula (A)

2. A process of Claim 1 wherein:

R and R independently are $C_{1-\delta}$ alkyl; and

20 X5 is carboxy or carboxy C1-6 alkyl.

3. A process of Claim 2 wherein:

M is C_{1-3} alkyl or allyl; and

 X_6 is hydrogen, C_{1-6} alkyl, or 3.4-methylenedioxy or phenyl.

4. A process of Claim 3 wherein:

25 R is ethyl; and

5

10

15

40

45

50

55

R' is methyl or ethyl.

5. A process of Claim 4 wherein:

R and Rt are ethyl:

M is n-propyl;

O X5 is 4-carboxymethyl; and

X₅ is 4-methyl.

6. A process of Claim 1 wherein:

Step (1) is in the presence of an alkali metal hydroxide:

Step (2) is in the presence of a tri(C1-5 alkyl)amine; and

the conversion of X₅ into X₅ is accomplished in the presence of a strong acid.



EUROPEAN SEARCH REPORT

DOCUMENTS CONSIDERED TO BE RELEVANT					EP 89200864.0
Category	Citation of document wit of relev	h indication, where a ant passages	ppropriate.	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CI.4)
x	EP - A1 - 0 199 (MERCK) * Claims 1,9		•	1-9	C 07 D 205/08 A 61 K 31/395
					•
					TECHNICAL FIELDS SEARCHED (Int. CI =) C 07 D 205/00
				•	
-					
	The present search report has b	een drawn up for all (claims		
VIENNA Date of complete VIENNA 14-07-19		etion of the search 989	JANISCH		
CATEGORY OF CITED DOCUMENTS X particularly relevant if taken alone Y particularly relevant if combined with another document of the same category A technological background O: non-written disclosure P intermediate document		T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons a: member of the same patent family, corresponding document			

THIS PAGE BLANK (USPTO)